

# Nanomanufacturing: Size-Dependent Lithium Solubility in Alloy Anodes

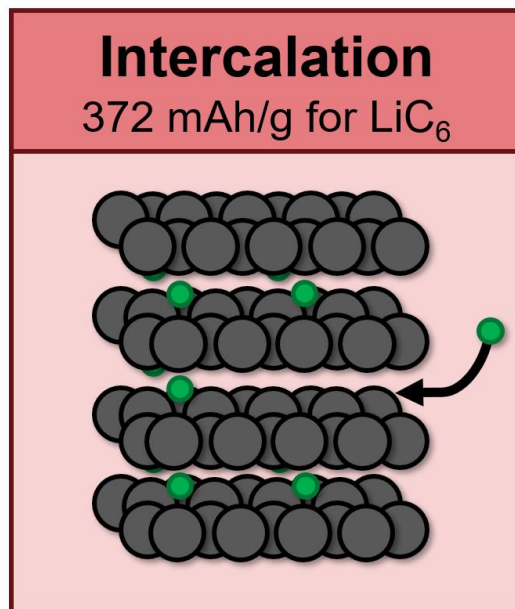
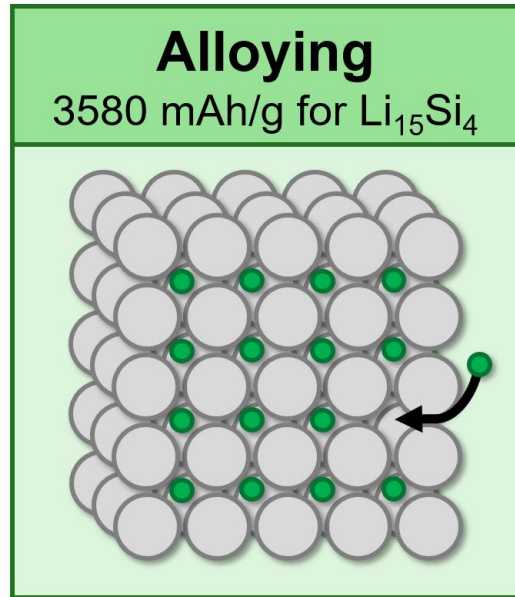
**Eric Detsi**, Associate Prof. Materials Science & Engineering

**Contribution:**

Dr. John Corsi (Detsi Group Alumnus)

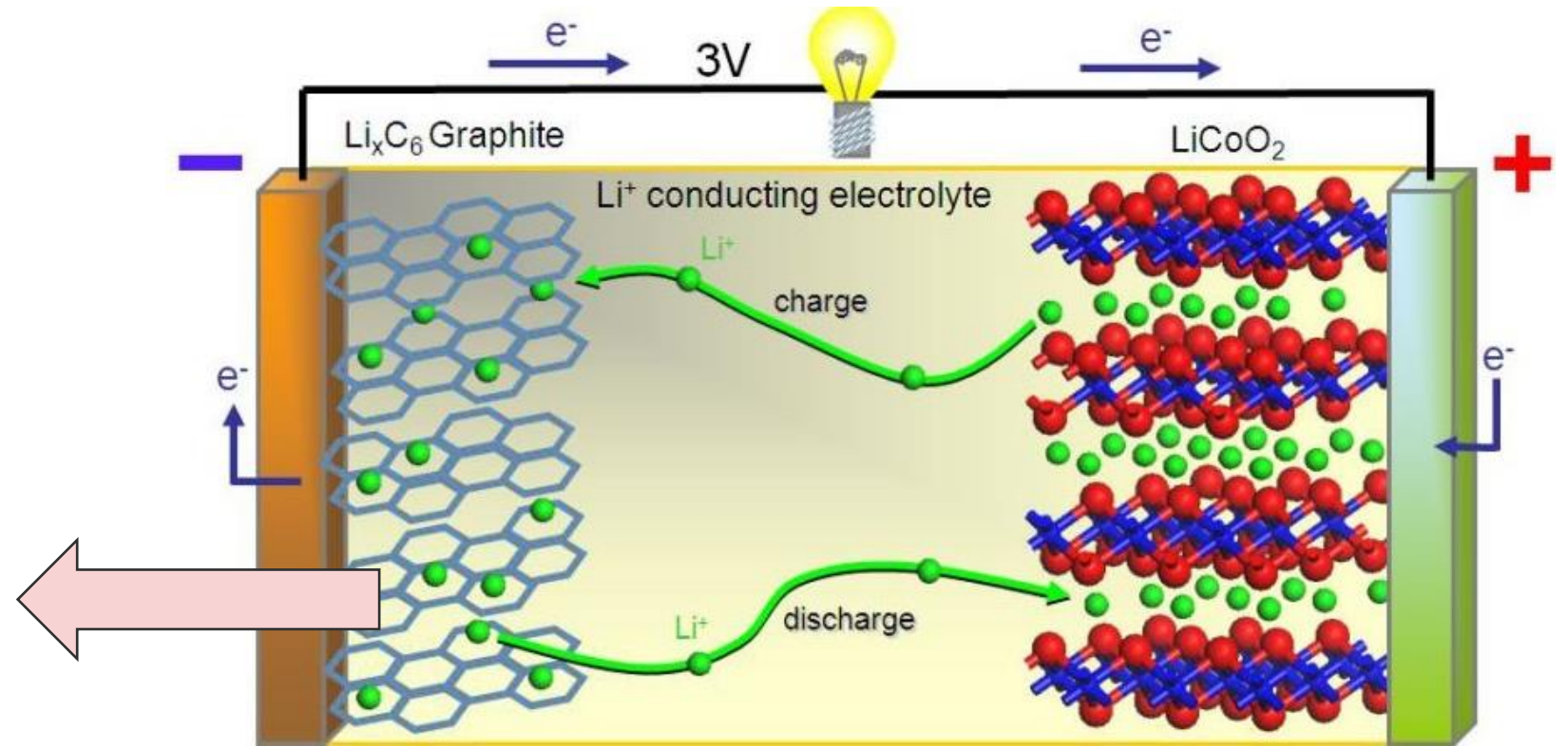
Dr. Sam Welborn (Detsi Group Alumnus)

# High-Capacity Alloy Anodes → Enhanced Energy Density



$$\Delta G = -q \cdot \Delta V$$

( $\Delta V = V_C - V_A$ )



# High-Capacity Alloy Anodes → Enhanced Energy Density

## Candidates Alloy Anodes for Li Storage

												He
												Ne
						B	372 756	N	O	F	Ne	
						993 1383	3579 2190	2596 2266	S	Cl	Ar	
<b>Mn</b>	<b>Fe</b>	<b>Co</b>	<b>Ni</b>	<b>Cu</b>	410 1511	769 1911	1384 2180	1073 2057	Se	Br	Kr	
Tc	<b>Ru</b>	Rh	Pd	248 1368	238 1159	1012 1980	960 1991	660 1889	Te	I	Xe	
Re	Os	Ir	Pt	510 2105	Hg	Tl	550 1906	385 1746	Po	At	Rn	

$\text{mAhg}^{-1}$   
 $\text{mAhcm}^{-3}$

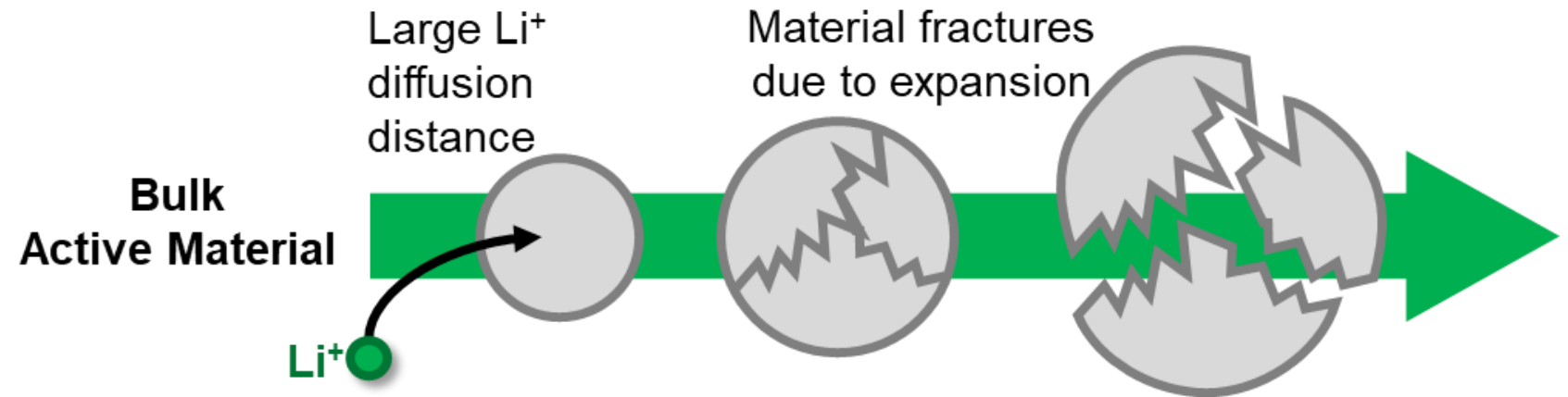
# High-Capacity Alloy Anodes → Enhanced Energy Density

## Problem: Macro-sized (Bulk) Alloy Anodes Fracture During Li Storage

### Critical particle size to avoid fracture

$$d_{cr} = \frac{\Gamma E}{Z \sigma_y^2}$$

- $d_{cr}$ : Critical size
- $\Gamma$ : Fracture energy
- $\sigma_y$ : Yield strength
- $E$ : Elastic modulus
- $Z$ : Material/crack param.



$$d_{cr}^{Sn} \leq 22 \text{ nm}$$

$$d_{cr}^{Al} \leq 345 \text{ nm}$$

# High-Capacity Alloy Anodes → Enhanced Energy Density

## Solution: Nano-Sized Alloy Anodes Should Not Fracture (?)

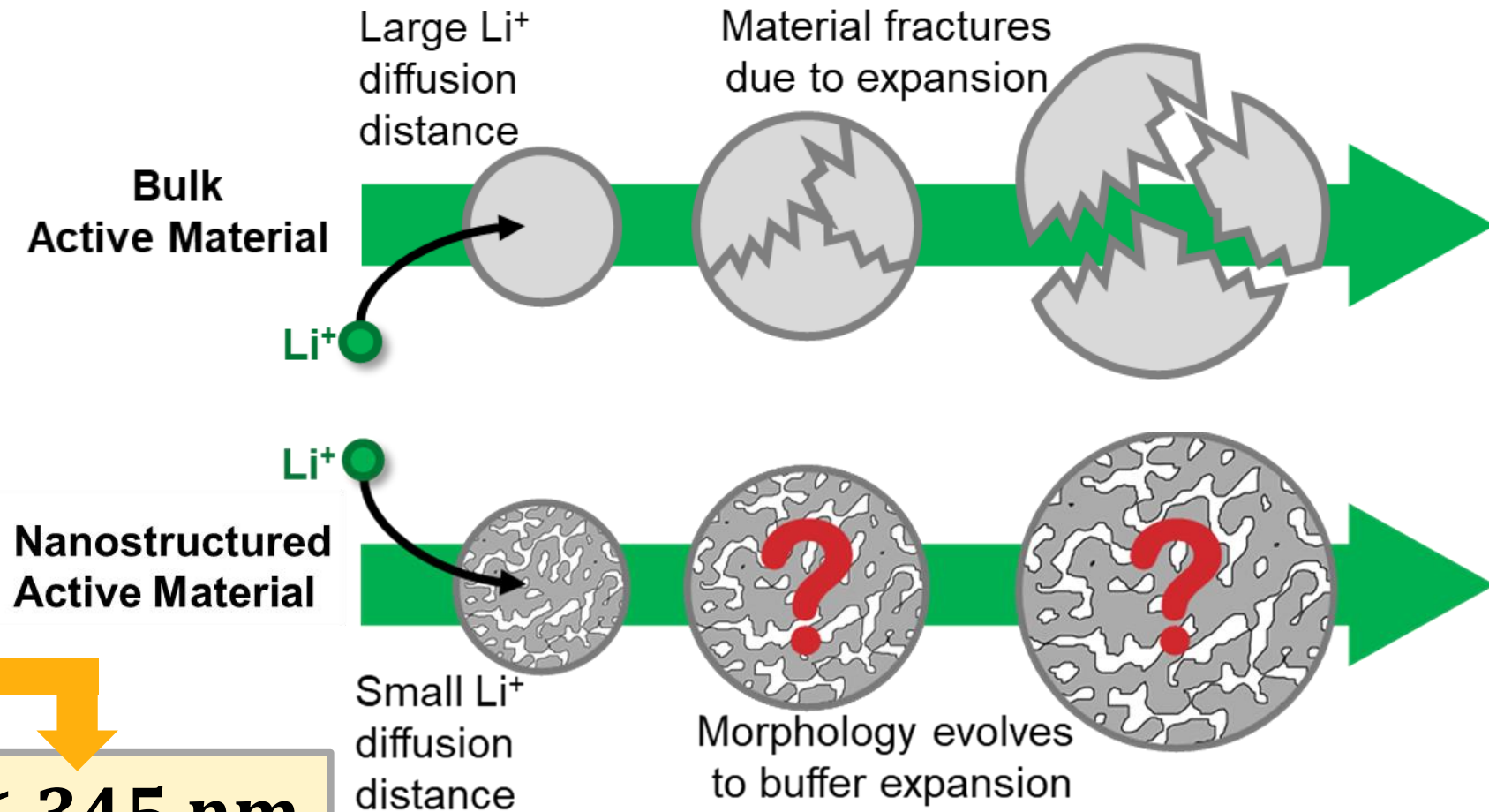
### Critical particle size to avoid fracture

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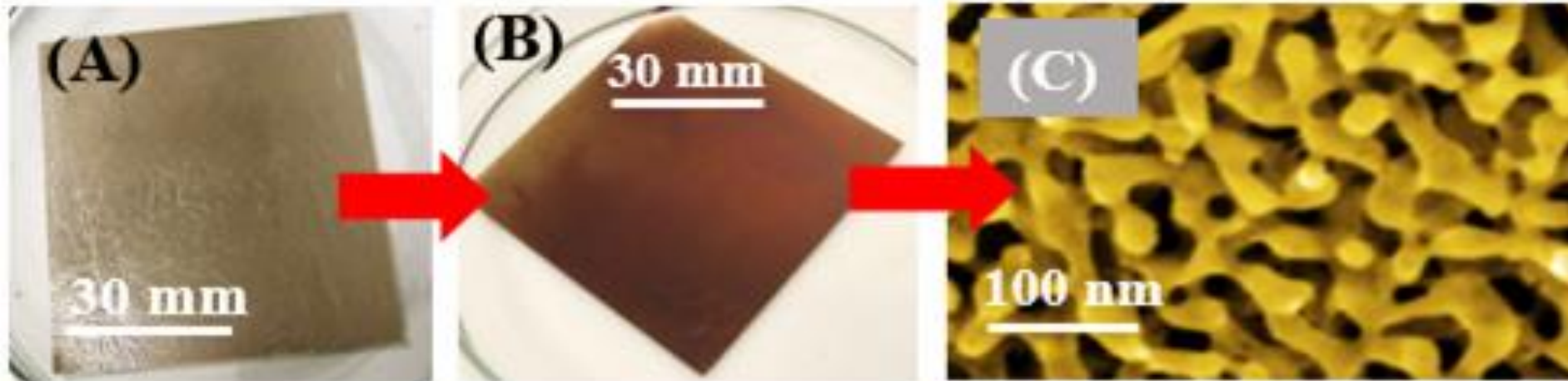
$$d_{cr}^{Sn} \leq 22 \text{ nm}$$

$$d_{cr}^{Al} \leq 345 \text{ nm}$$



# Creating Nano-Sized Alloy Anodes by Dealloying

## Example 1: Nanoporous Gold (NP-Au)



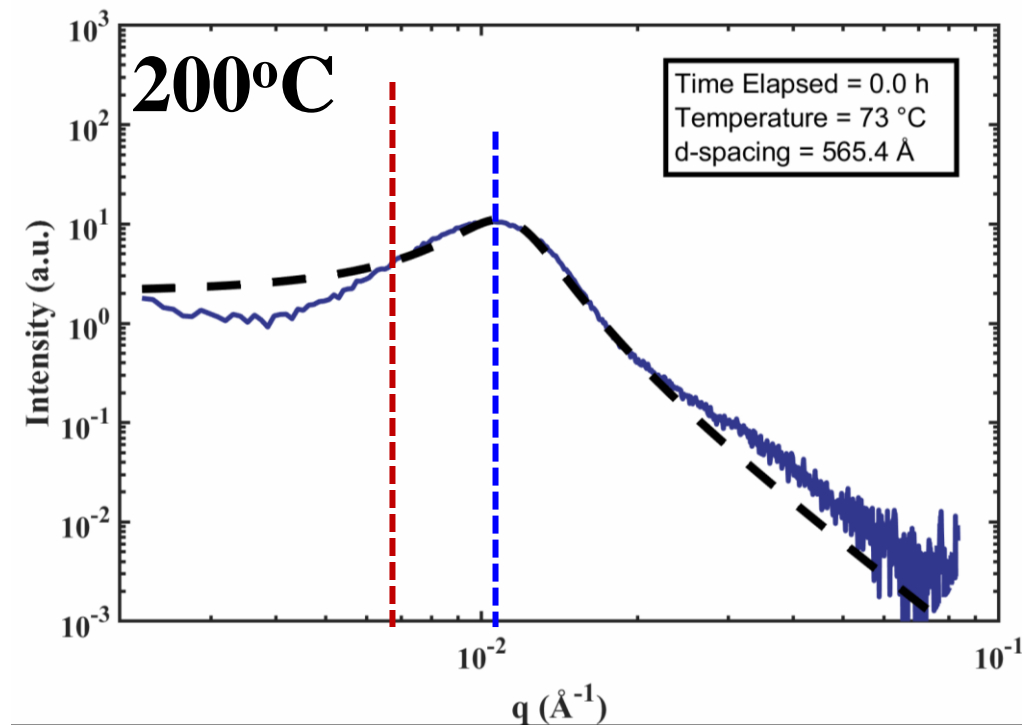
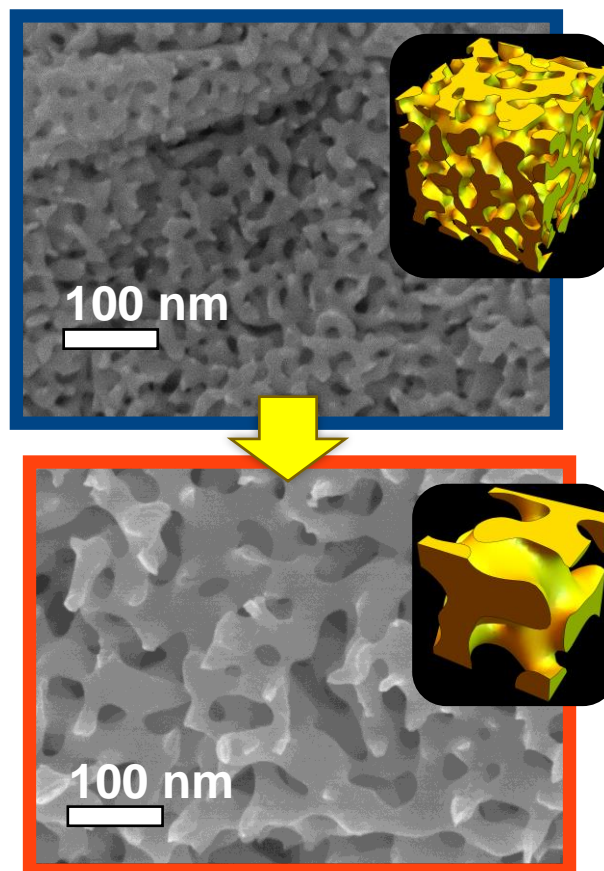
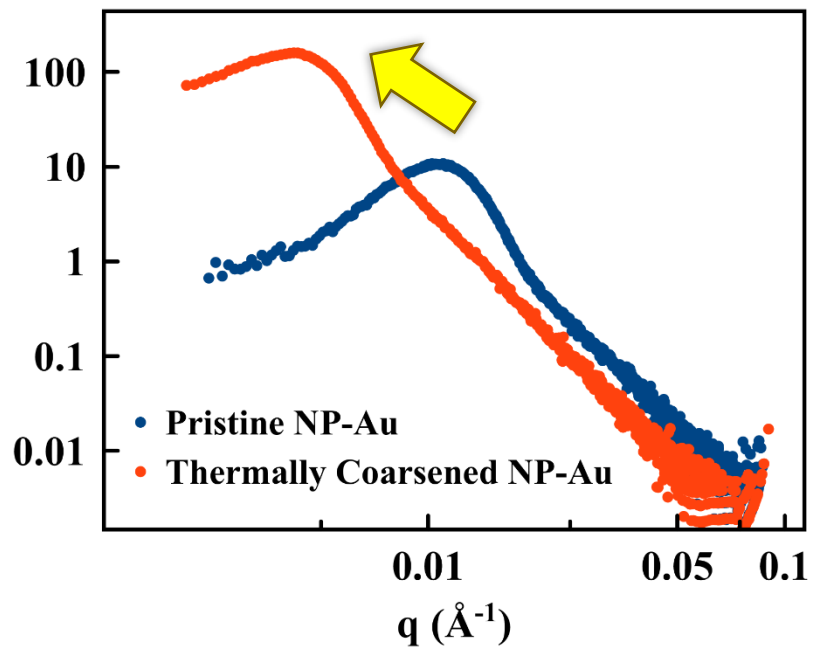
**Precursor alloy film**  
( $\text{Au}_{35}\text{Ag}_{65}$  at. %)

**Nanoporous-Au film**  
(naked eye)

**Nanoporous-Au**  
(electron microscopy)

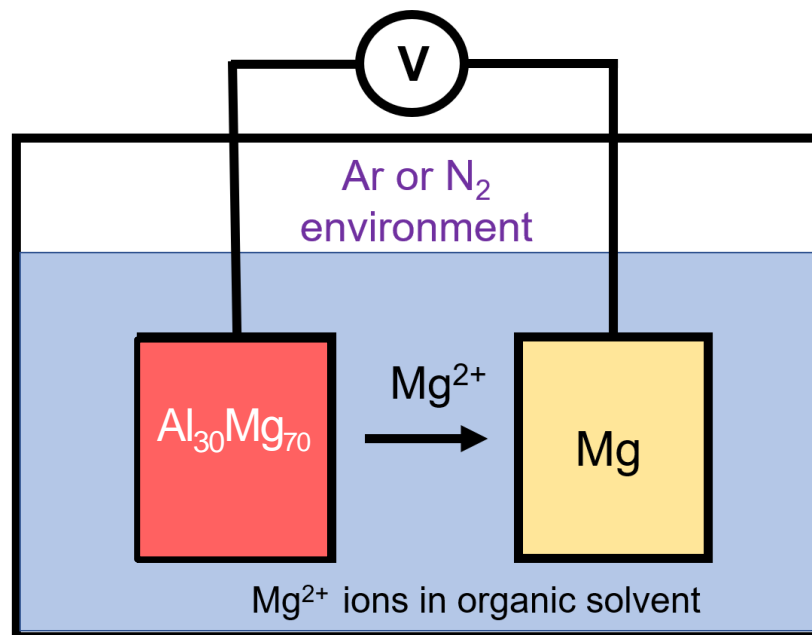
# Creating Nano-Sized Alloy Anodes by Dealloying

## Example 1: Nanoporous Gold (NP-Au)



# Creating Nano-Sized Alloy Anodes by Dealloying

## Example 2: Ultrafine Nanoporous Aluminum (NP-Al)



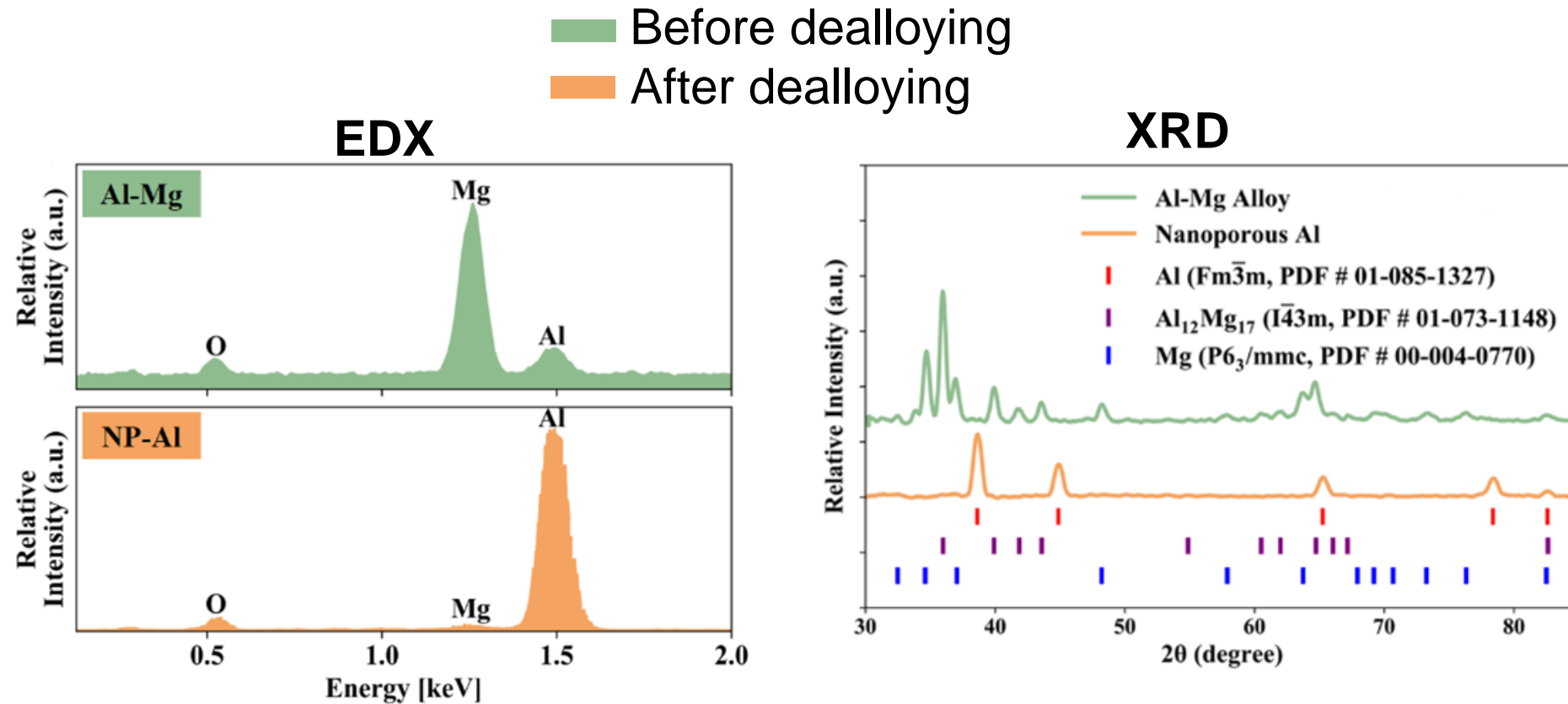
Working Electrode:  
 $\text{Mg} \rightarrow \text{Mg}^{2+} + 2\text{e}^-$

Counter Electrode:  
 $\text{Mg}^{2+} + 2\text{e}^- \rightarrow \text{Mg}$



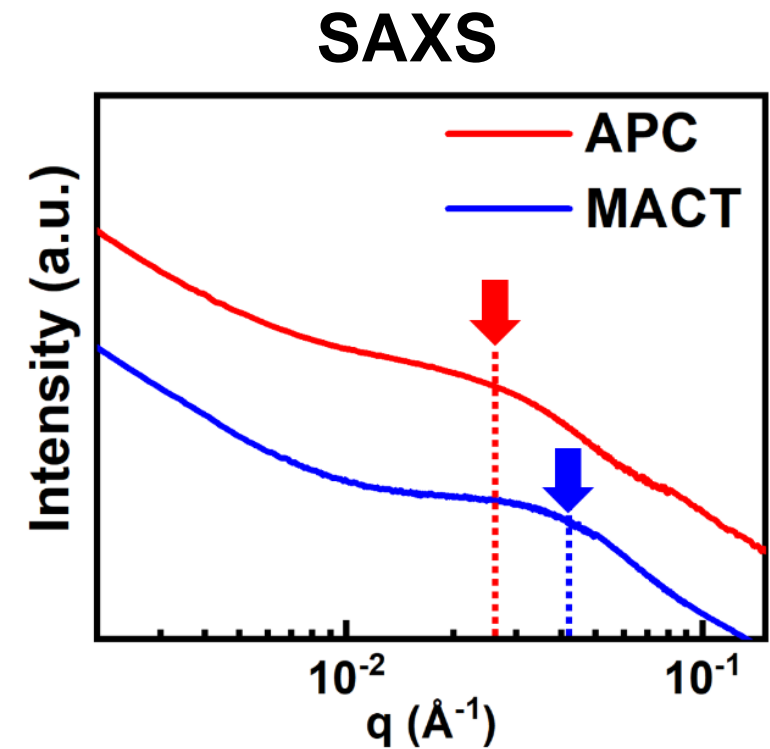
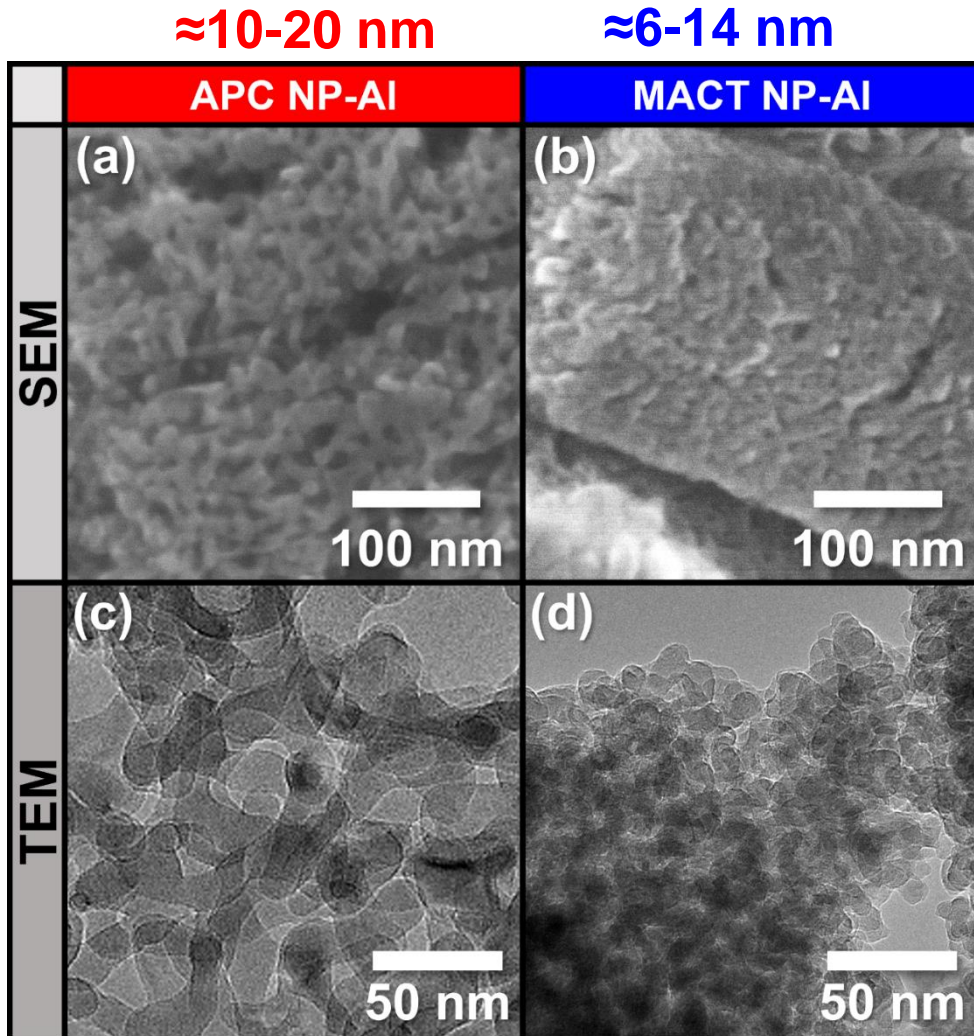
# Creating Nano-Sized Alloy Anodes by Dealloying

## Example 2: Ultrafine Nanoporous Aluminum (NP-Al)



# Creating Nano-Sized Alloy Anodes by Dealloying

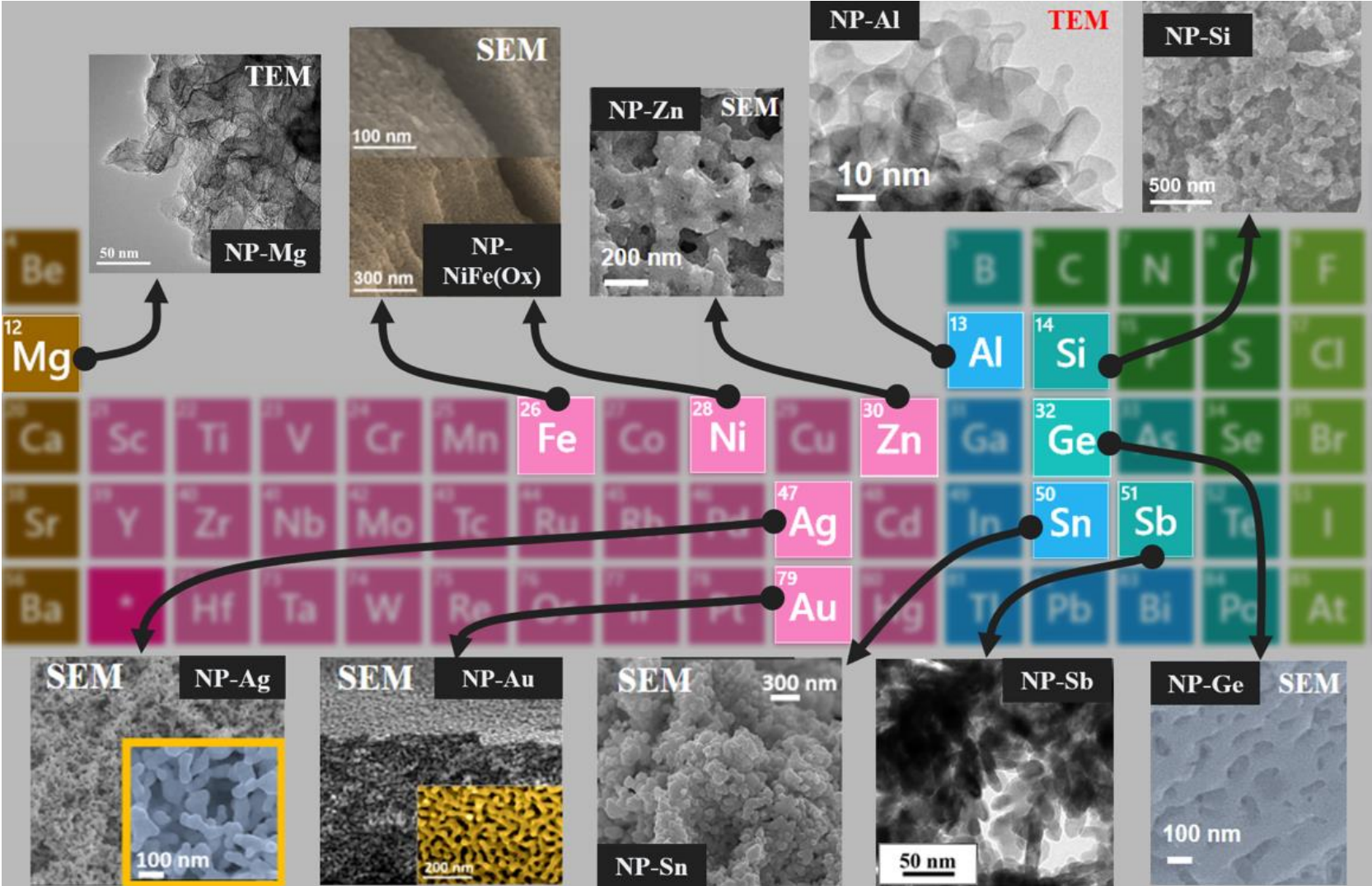
## Example 2: Ultrafine Nanoporous Aluminum (NP-Al)



T. Lee, ..., E. Detsi; *Scripta Materialia* 221, 114959 (2022)

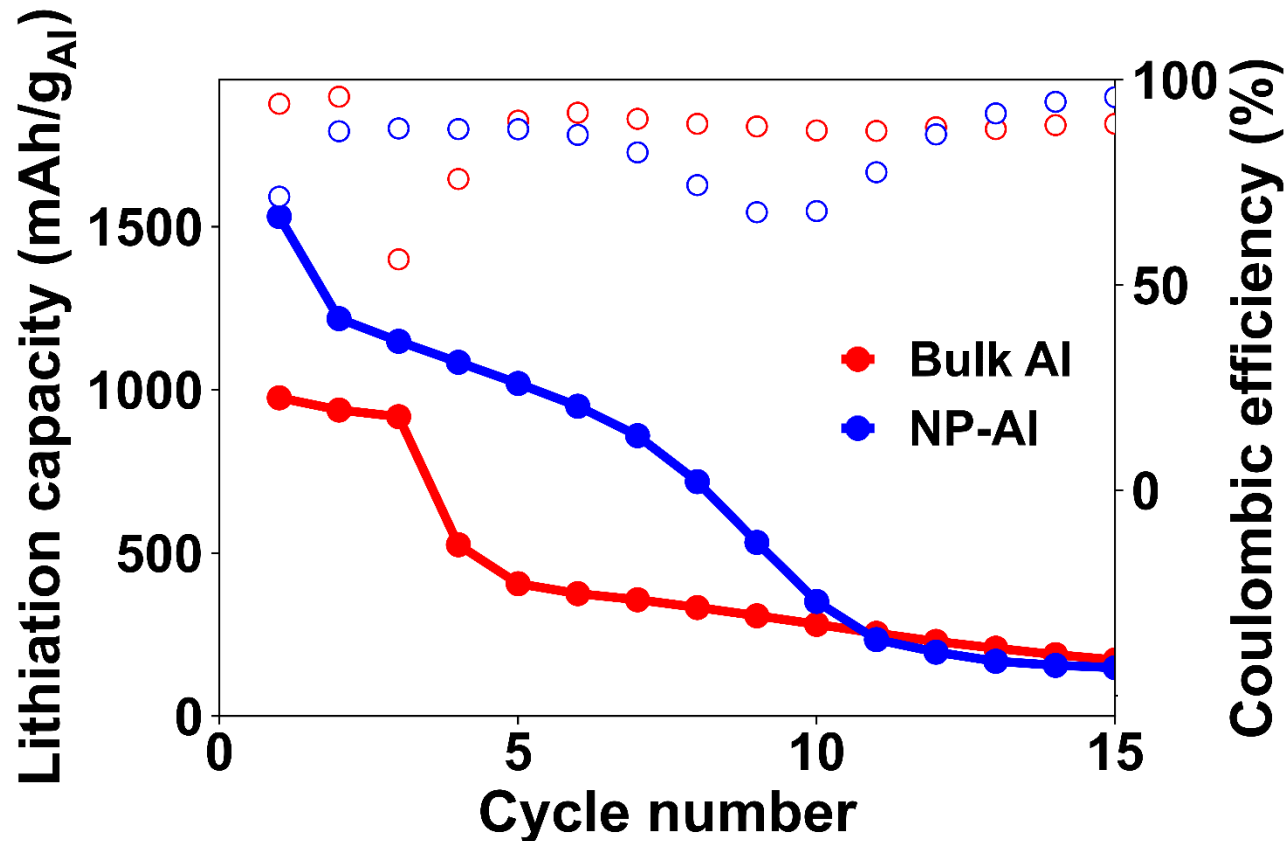
# Creating Nano-Sized Alloy Anodes by Dealloying

## More Examples of Dealloyed Nanoporous Materials



# Performance: Macro-Sized vs. Nano-Sized Alloy Anode

## Case Study: Bulk Aluminum vs. Nanoporous Aluminum



**NP-Al anode performs better than bulk Al anode**

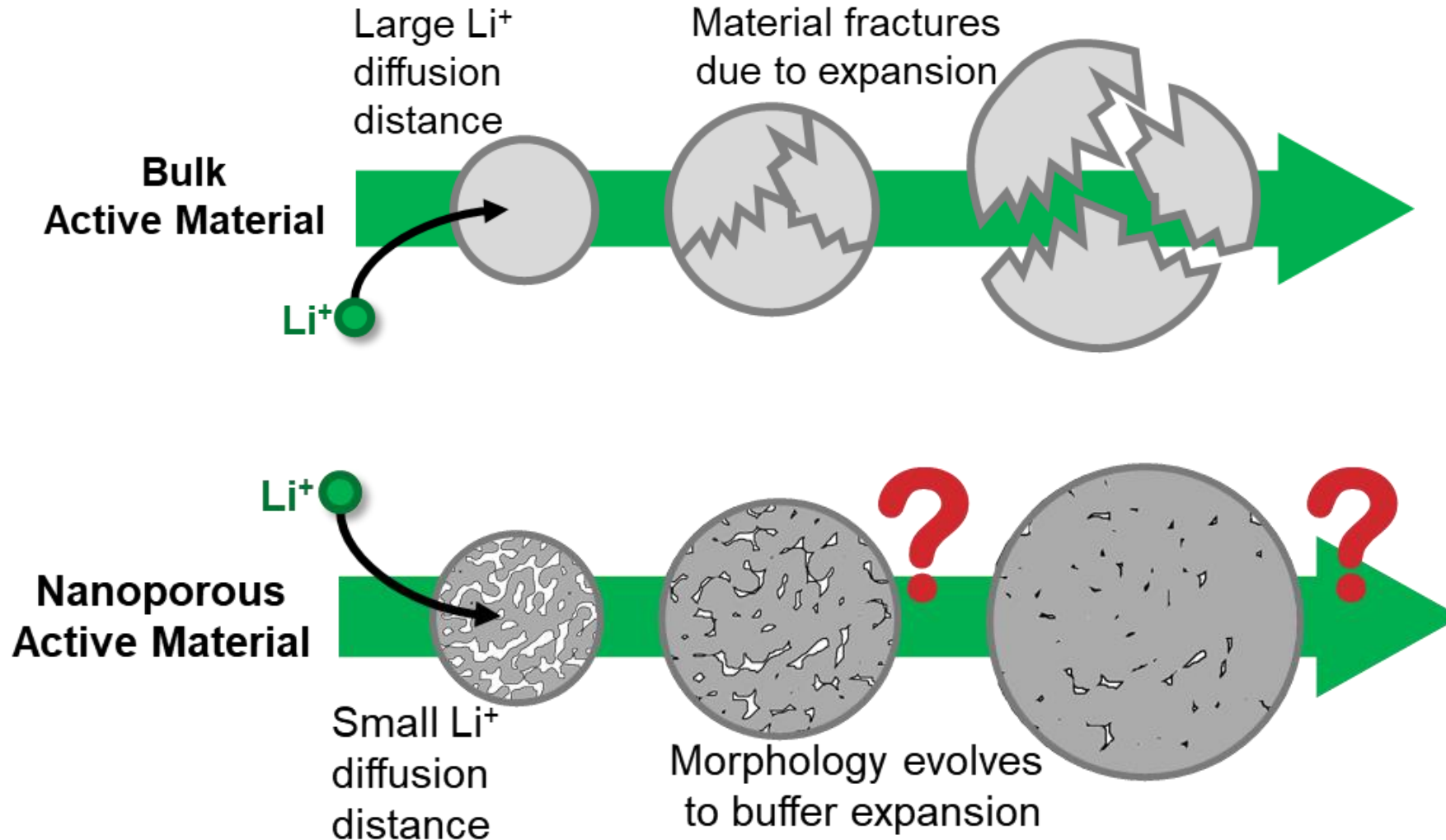
**NP-Al anode still fails after only a few cycles**

$$d^{\text{NP-Al}} < d_{\text{cr}}^{\text{Al}}$$

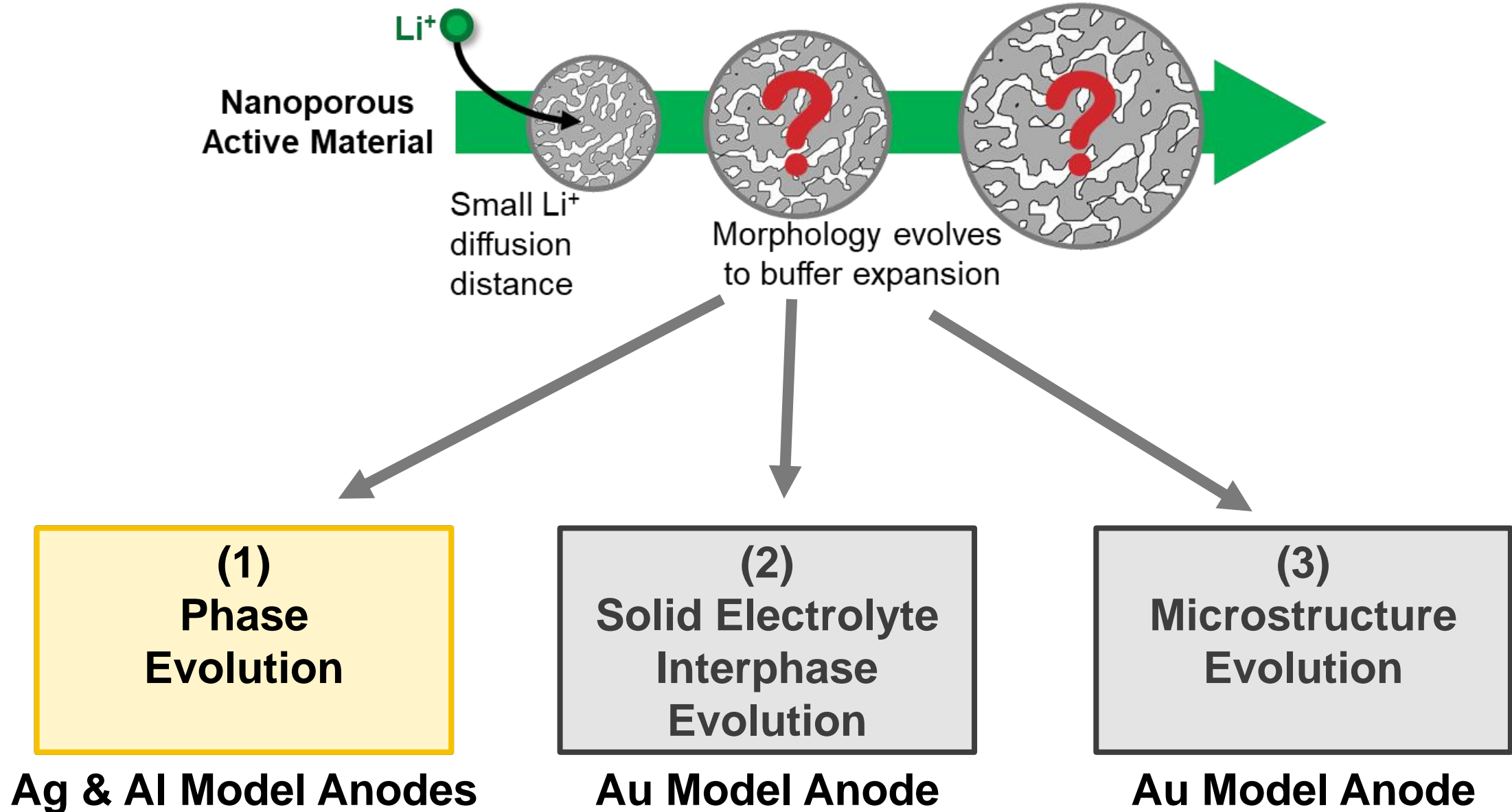
$$15 \text{ nm} \ll 345 \text{ nm}$$

# Performance: Macro-Sized vs. Nano-Sized Alloy Anode

## Why & How Do Nano-Sized Alloy Anodes Fail?



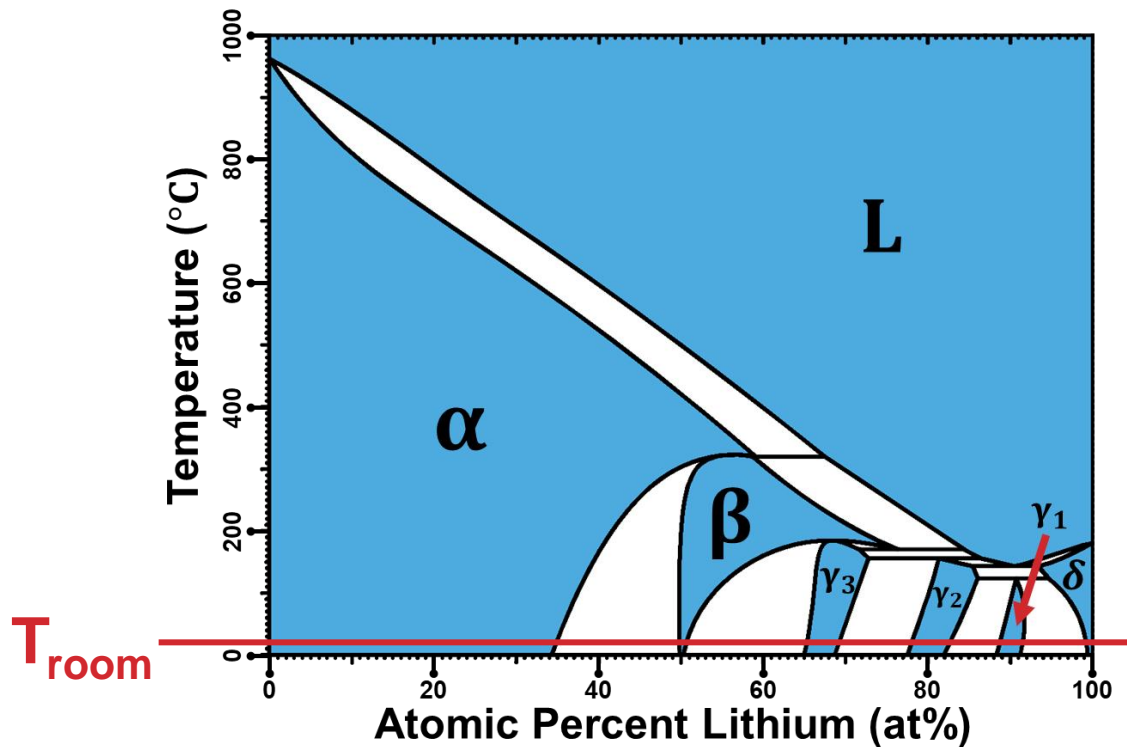
# Morphology Evolution During Li Storage in Nanoporous Alloy Anodes



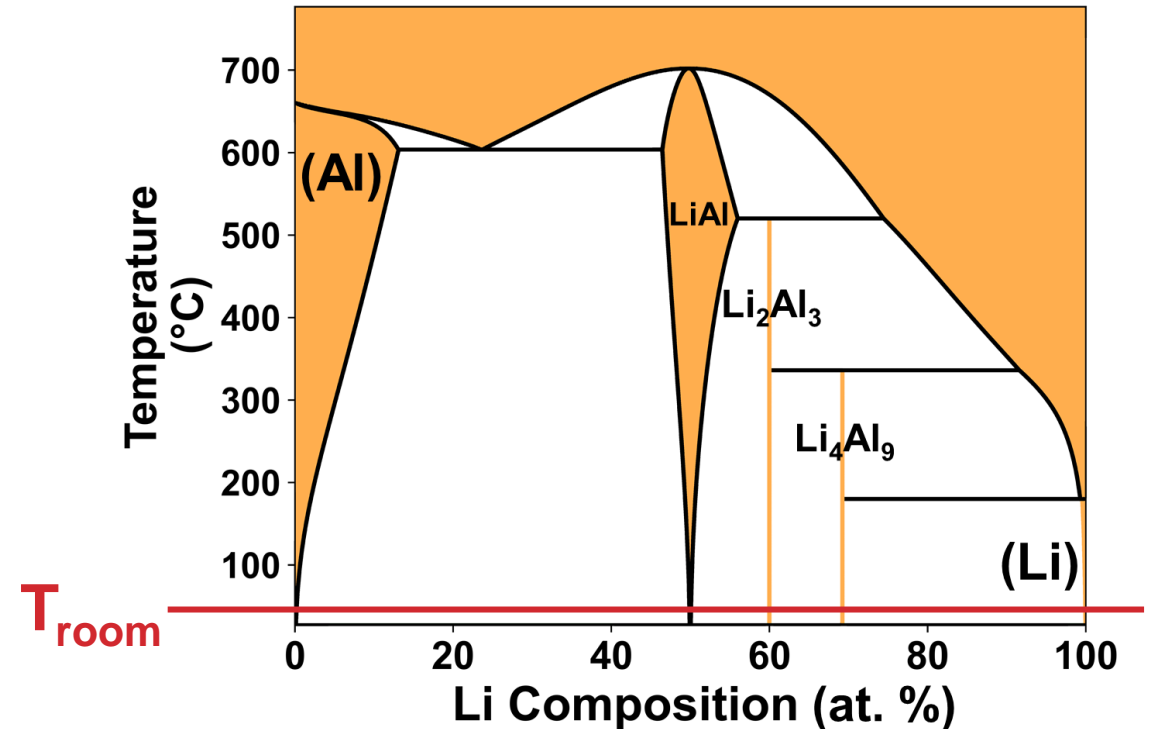
# (1) Phase Evolution

## Phase Diagrams for Ag and Al as Li-ion Battery Anodes

### Ag – Li Phase Diagram

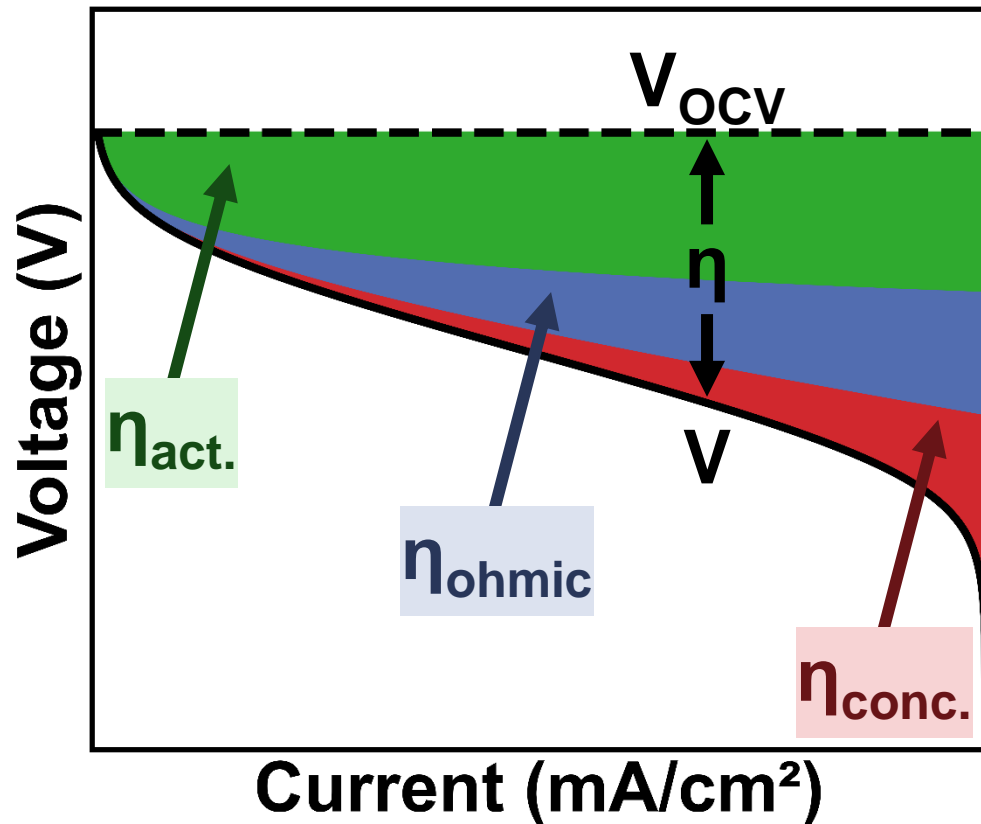


### Al – Li Phase Diagram



# (1) Phase Evolution

## Background (1): Voltage Profile



$$V = V_{ocv} - \eta$$

Equilibrium thermodynamic phase stability

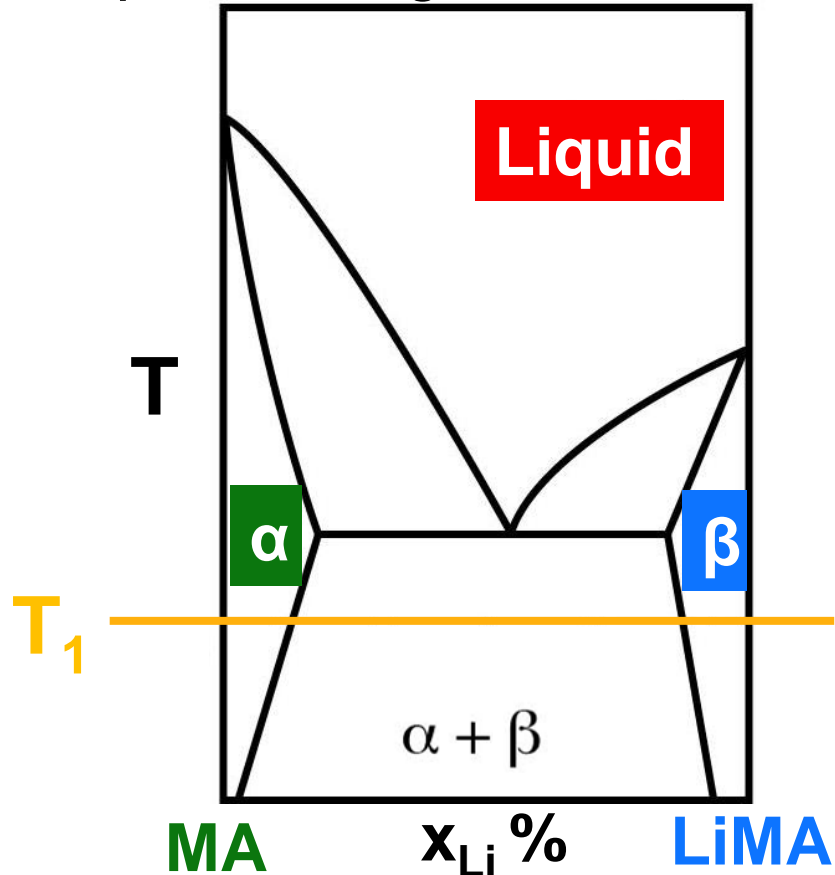
Kinetic effects



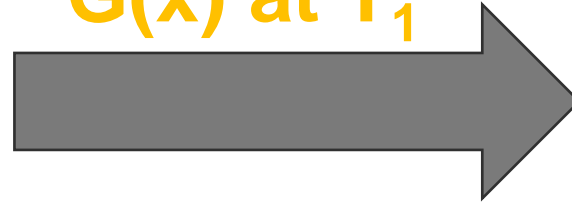
# (1) Phase Evolution

## Background (2): From Phase Diagram to Gibbs Free Energy and Voltage Profile

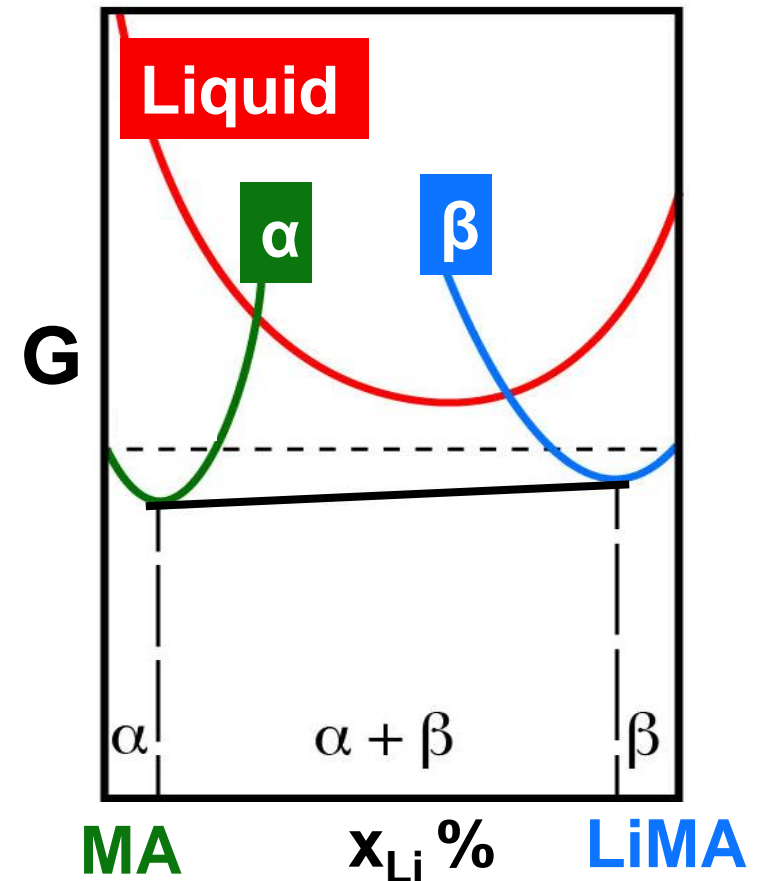
Hypothetical equilibrium phase diagram



$G(x)$  at  $T_1$

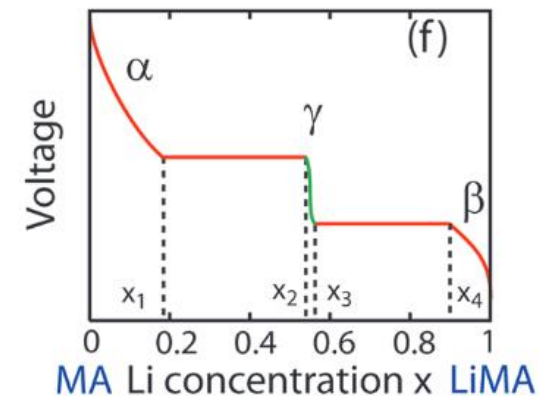
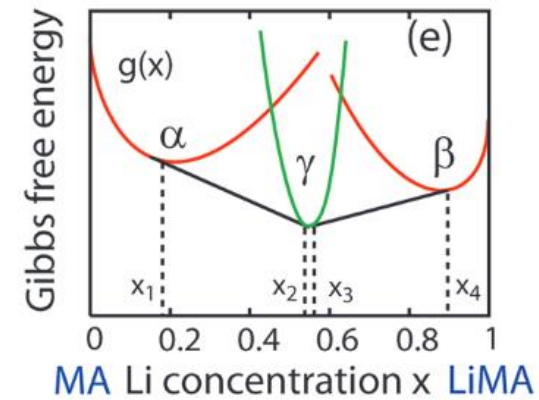
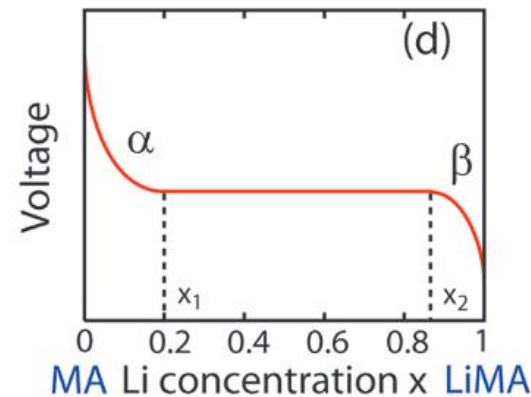
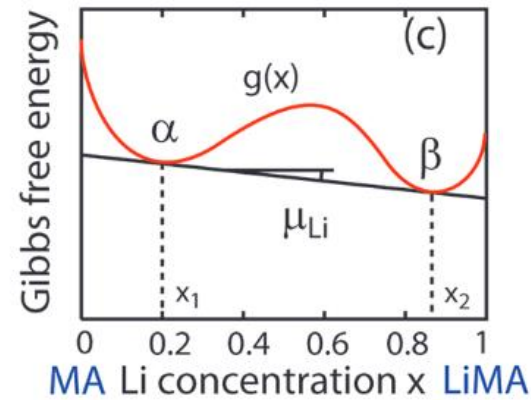
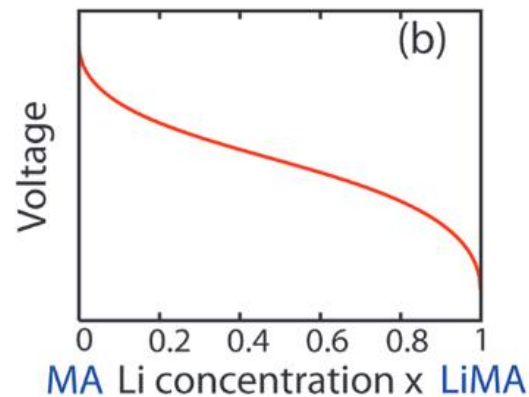
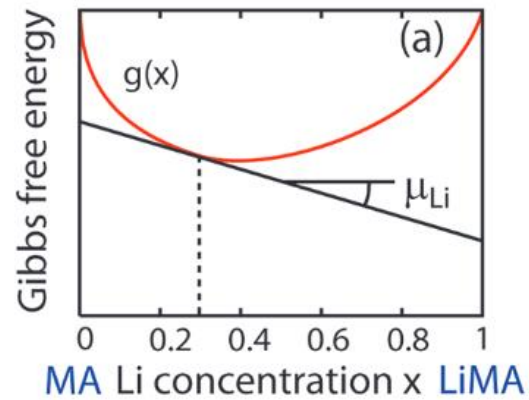


Corresponding free energy curves at  $T_1$



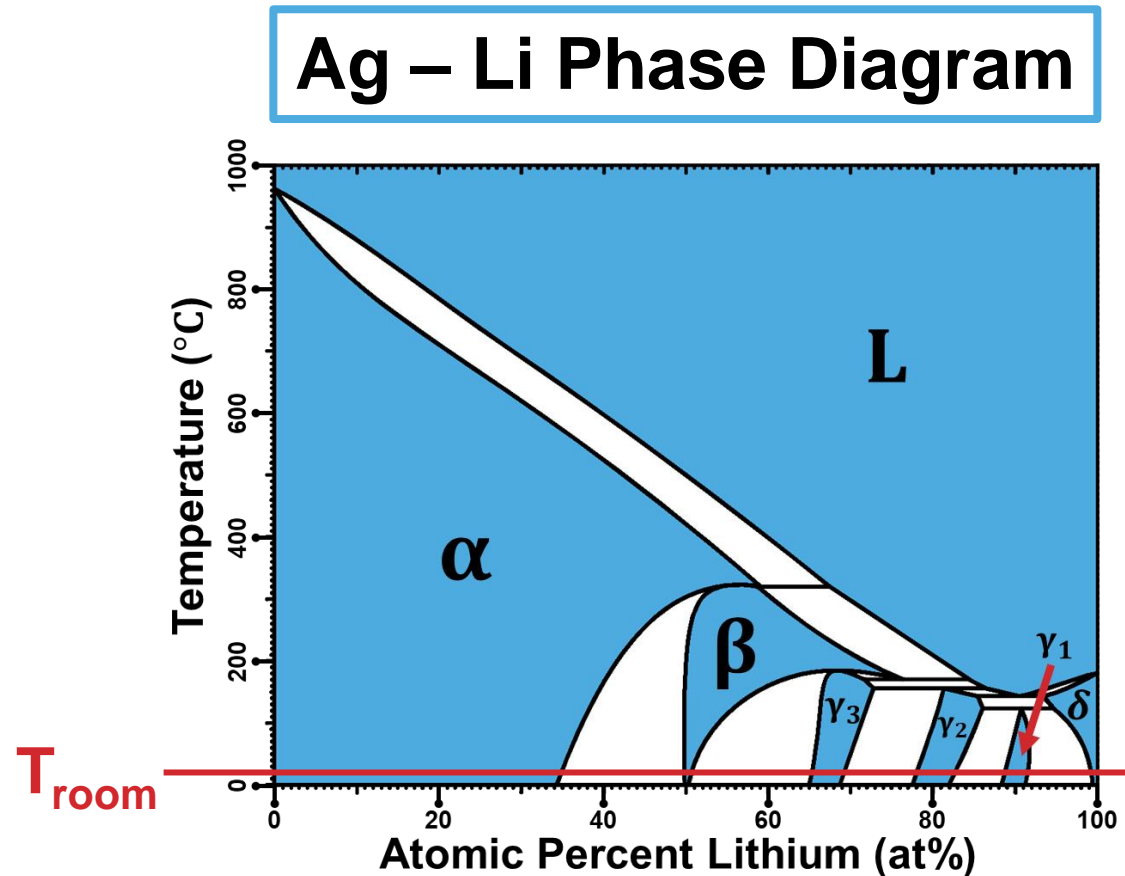
# (1) Phase Evolution

## Background (2): From Phase Diagram to Gibbs Free Energy and Voltage Profile



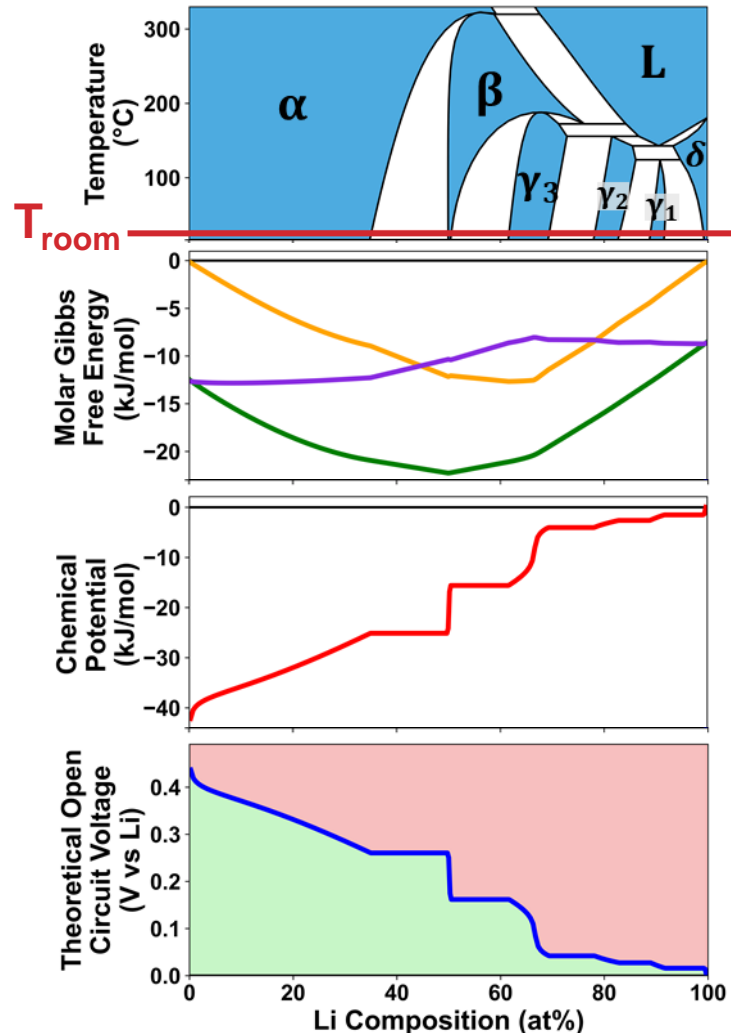
# (1) Phase Evolution

## Case 1: Ag-Li System



# (1) Phase Evolution

## Theoretical OCV Profile of Ag-Li System



(1) **FactSage™**

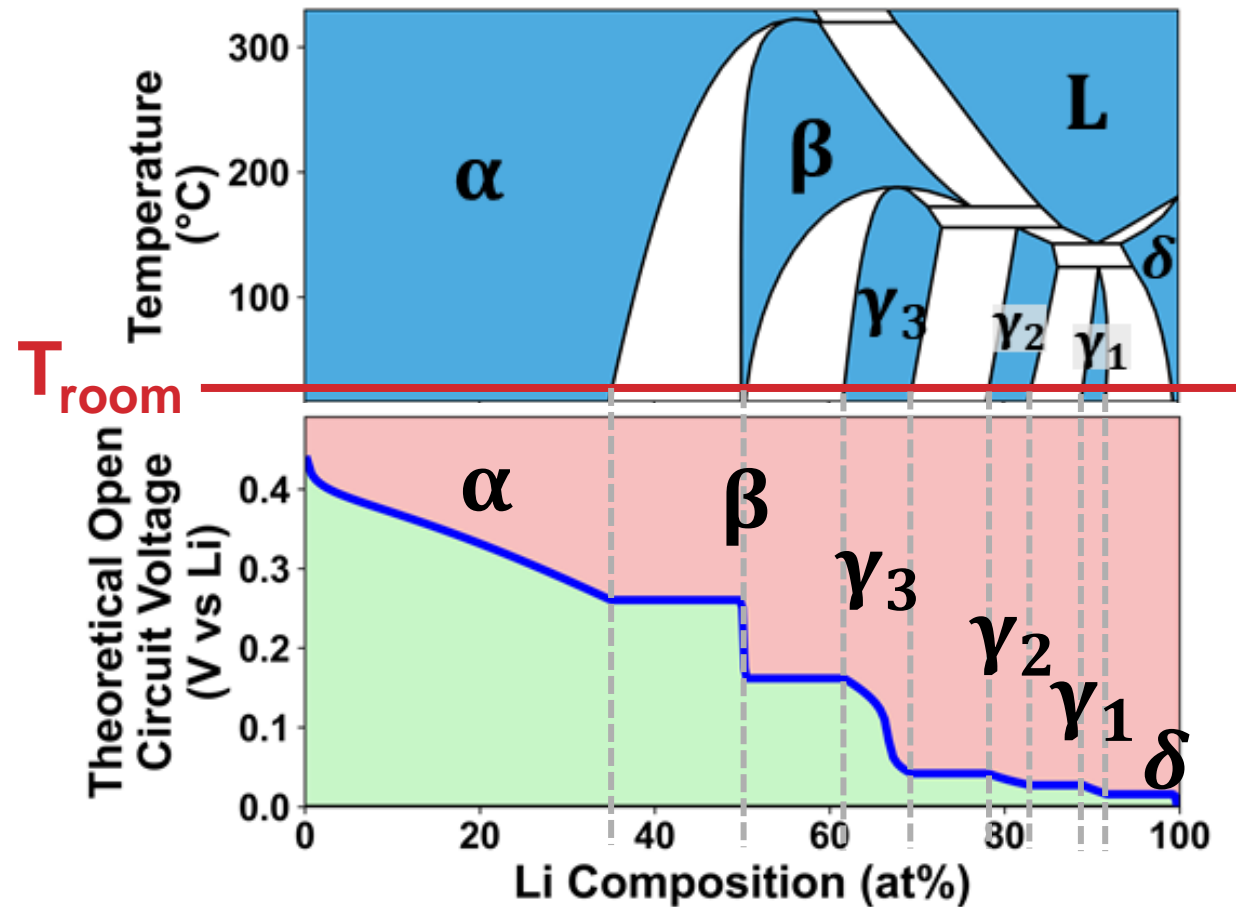
$$(2) \quad G(\mathbf{x}) = H(\mathbf{x}) - TS(\mathbf{x})$$

$$(3) \quad \Delta\mu_{\text{Li}} = \mu_{\text{Li}}(\mathbf{x}) - \mu_{\text{Li}}^0$$

$$(4) \quad E_{\text{OCV}} = -\frac{\Delta\mu_{\text{Li}}}{zF}$$

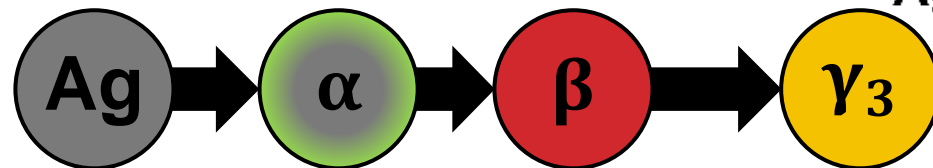
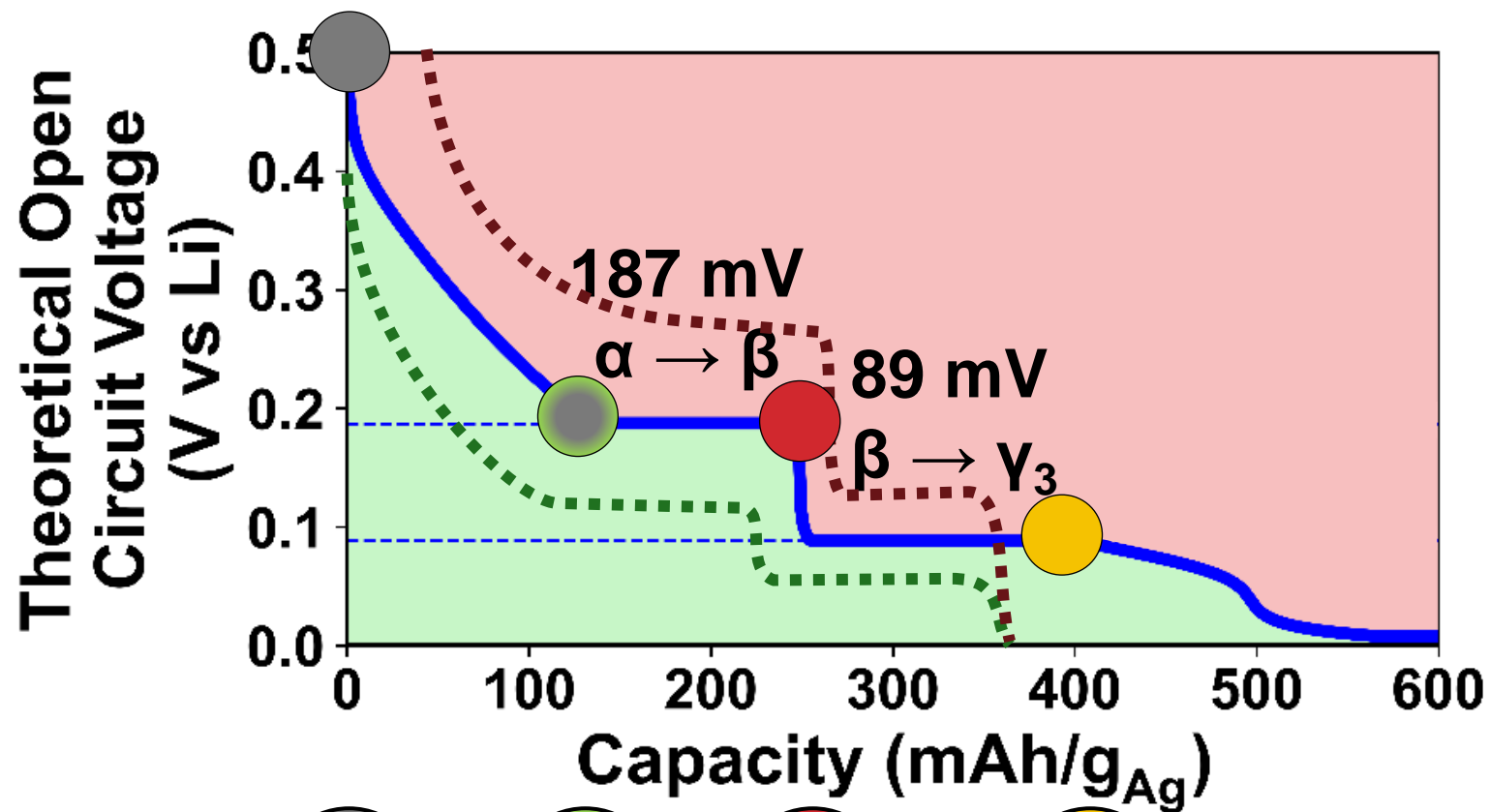
# (1) Phase Evolution

## Theoretical OCV Profile of Ag-Li System



# (1) Phase Evolution

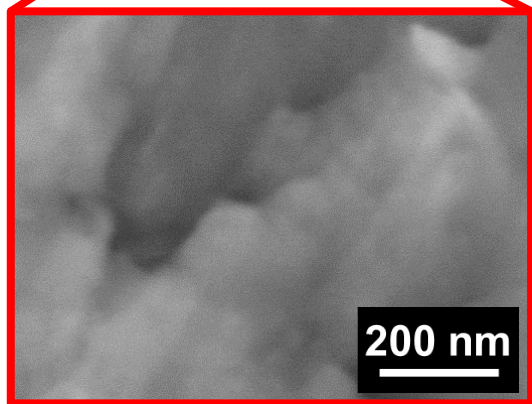
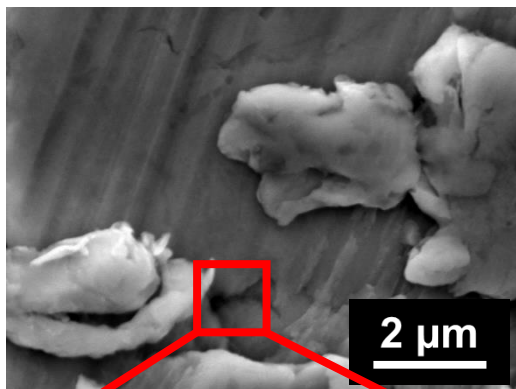
## Theoretical OCV Profile of Ag-Li System



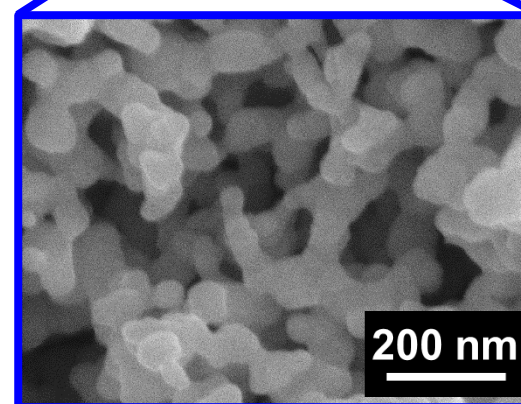
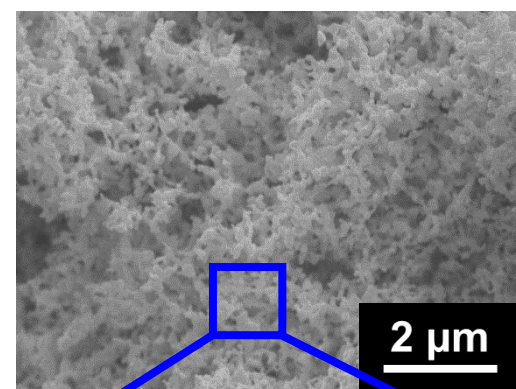
# (1) Phase Evolution

## Two Different Ag Morphologies Considered

**Bulk**

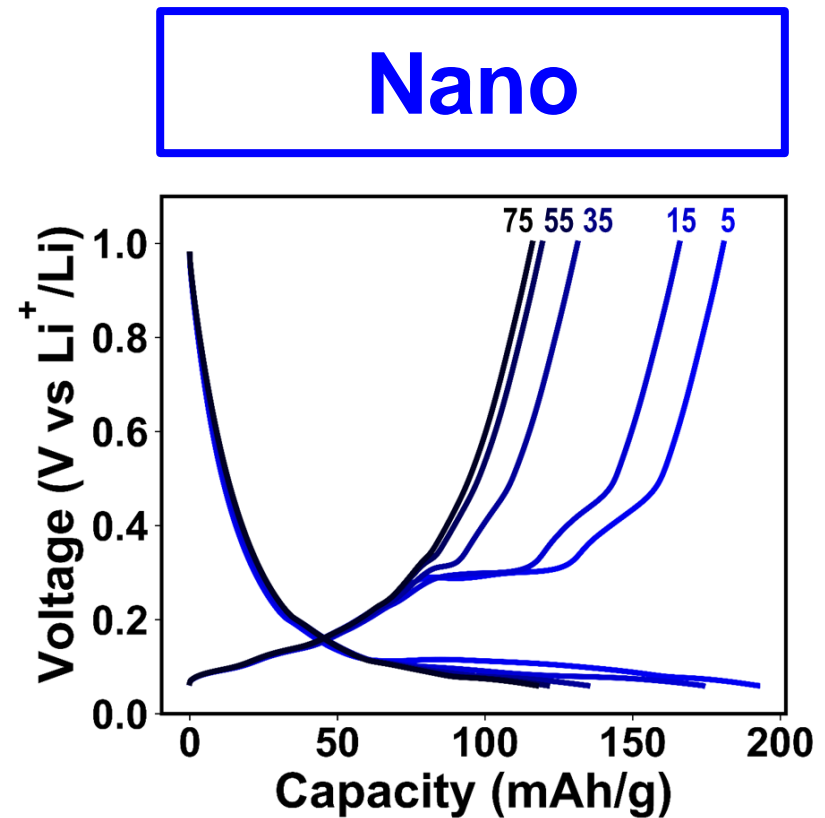
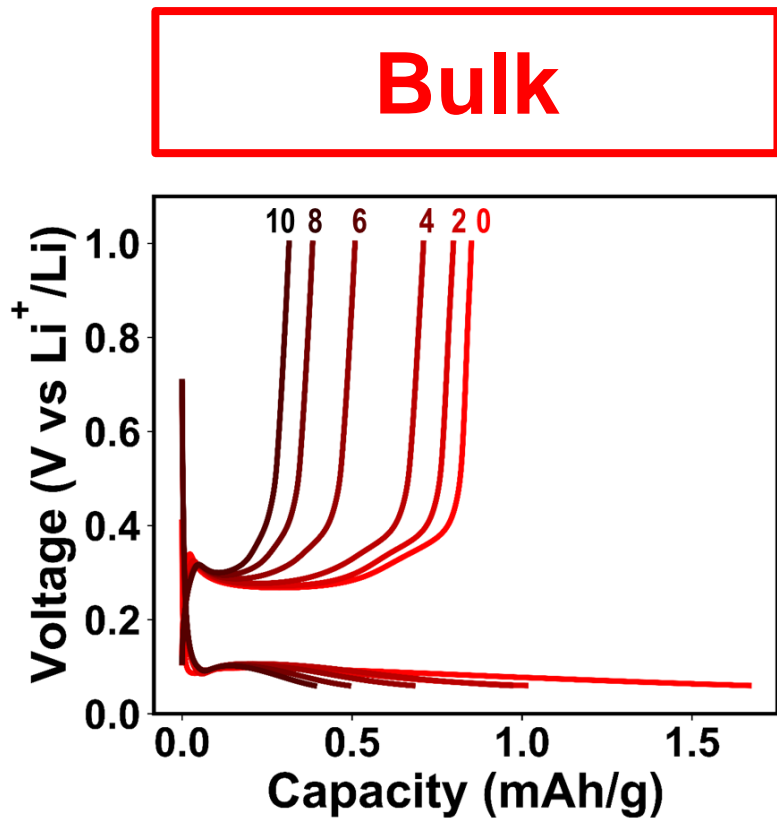


**Nano**



# (1) Phase Evolution

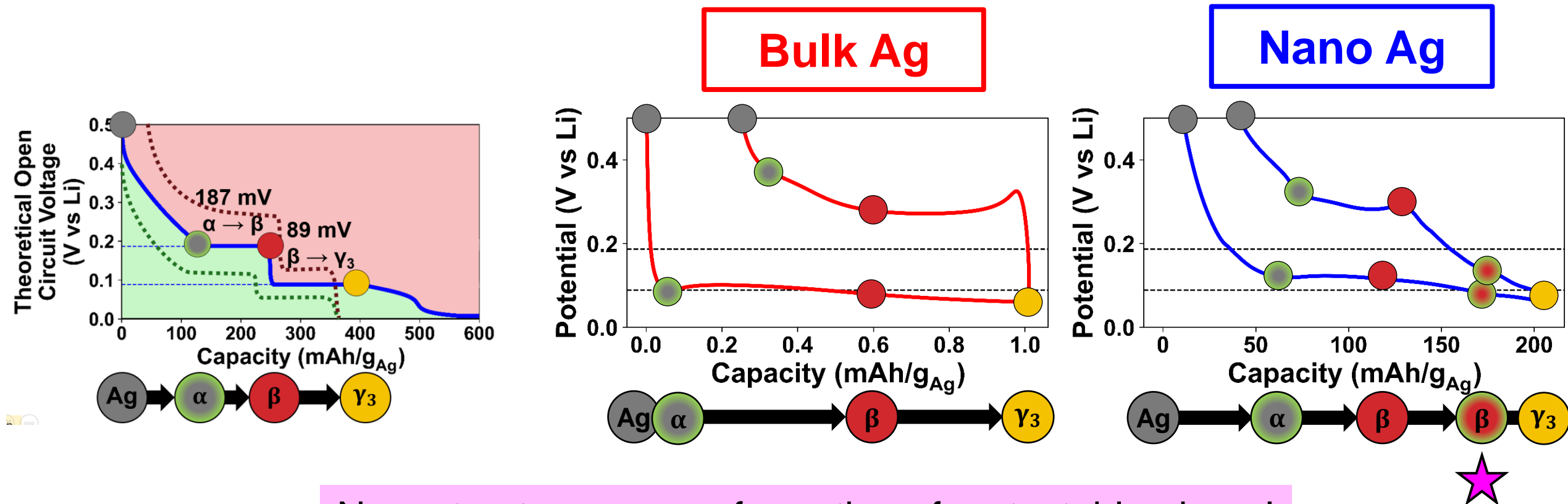
## Bulk vs Nanoporous Ag Cycling Performance





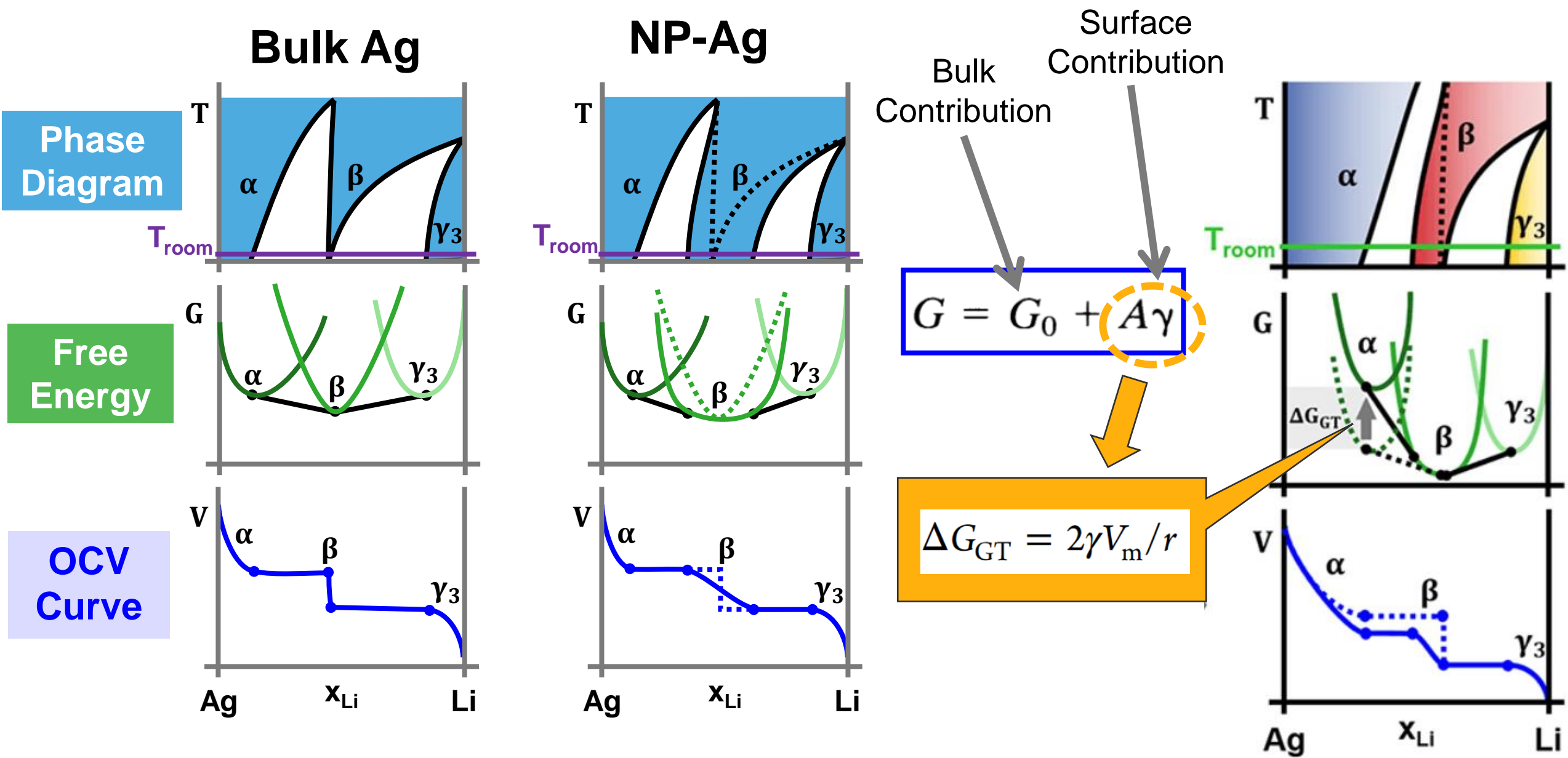
# (1) Phase Evolution

## Experimental Voltage Profile of Ag-Li System



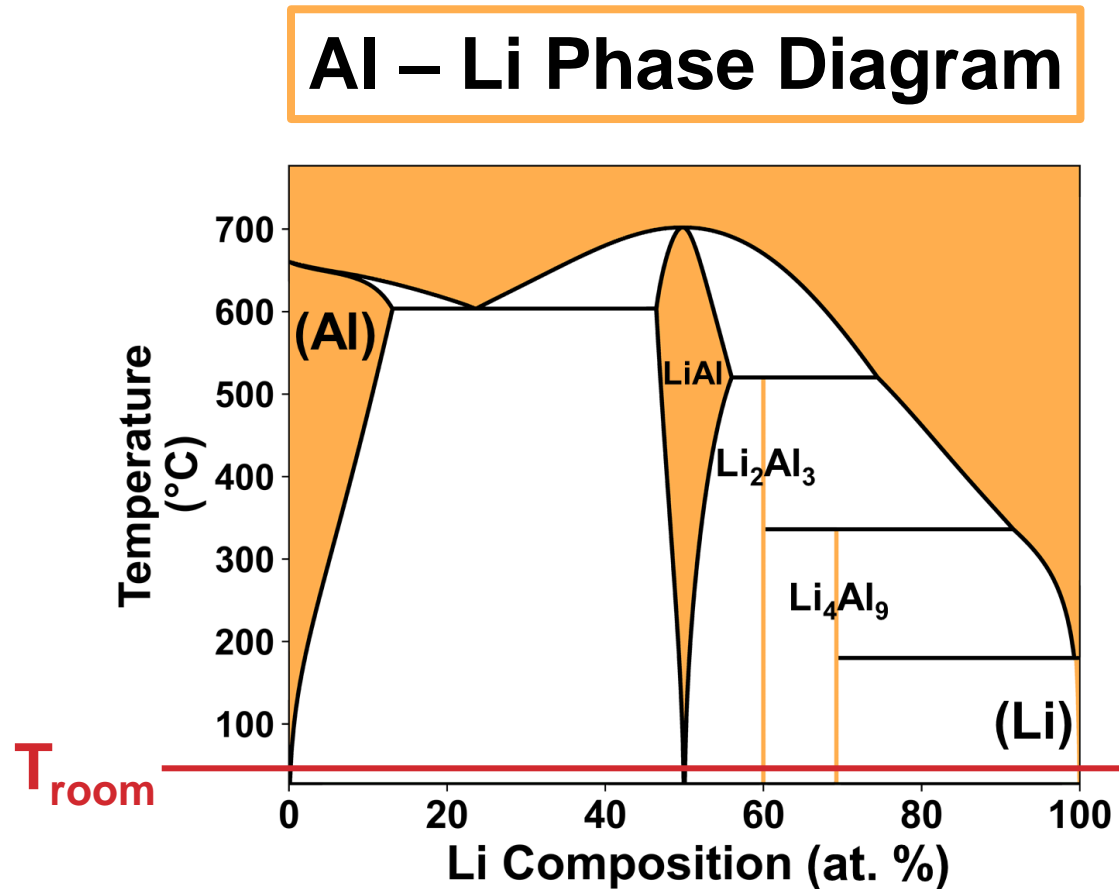
Nanostructure causes formation of metastable phase!

# (1) Phase Evolution



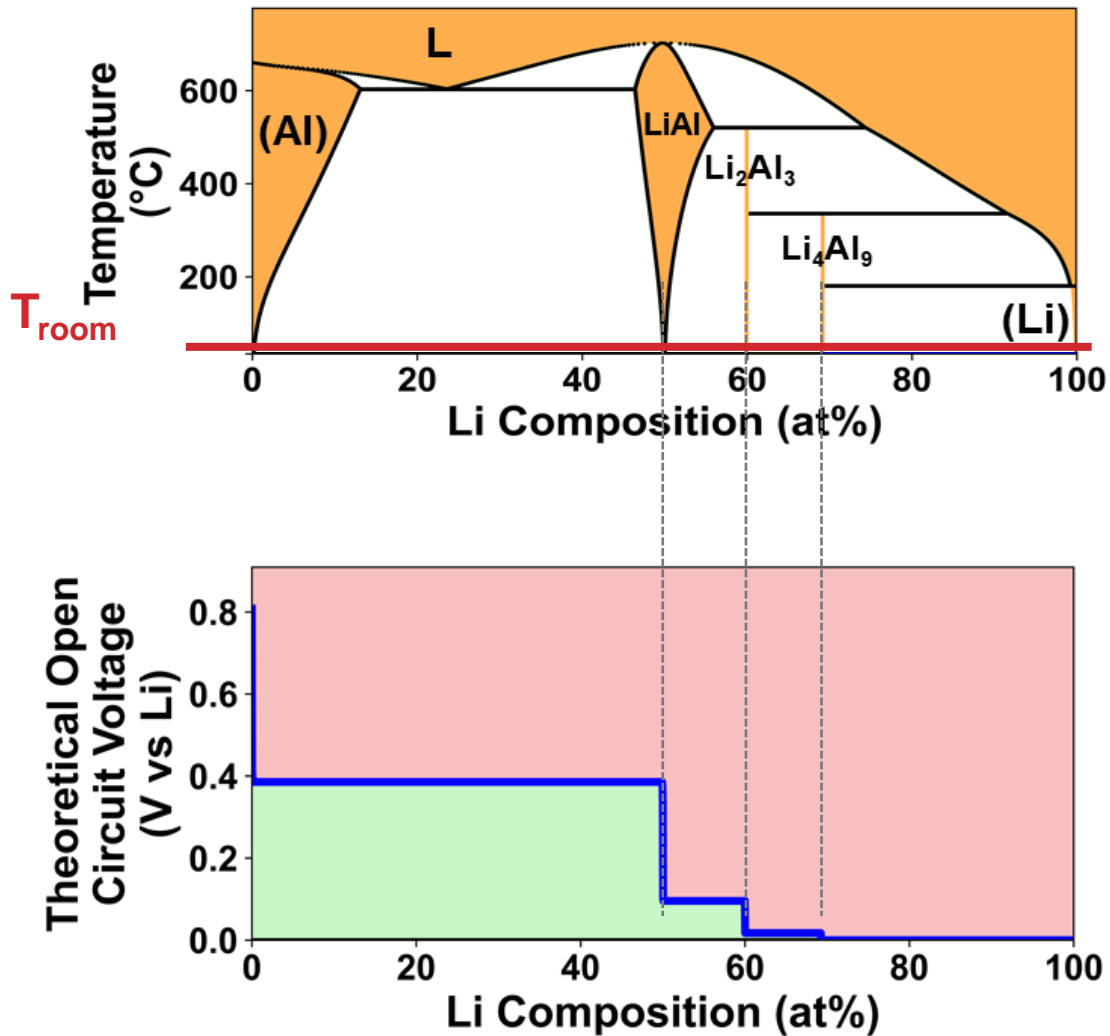
# (1) Phase Evolution

## Case 2: Al-Li System



# (1) Phase Evolution

## Theoretical OCV Profile of Al-Li System

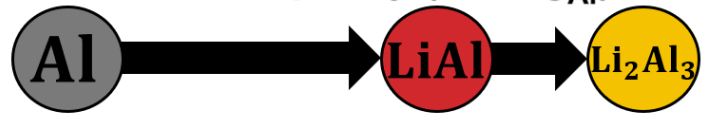
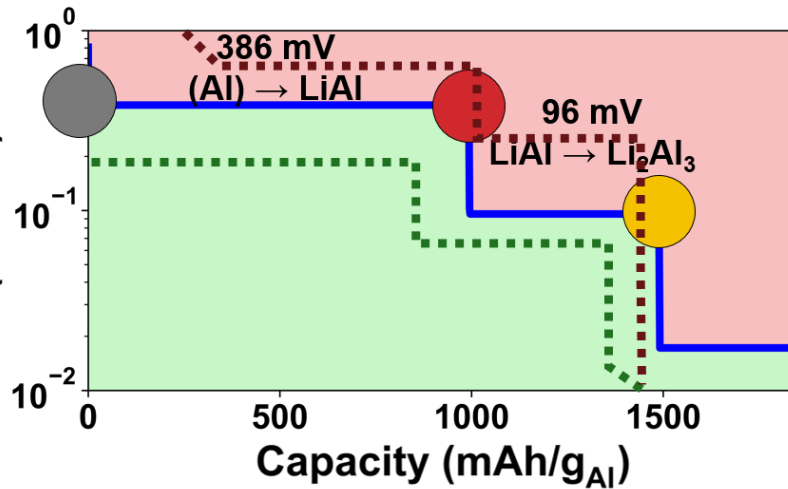


# (1) Phase Evolution

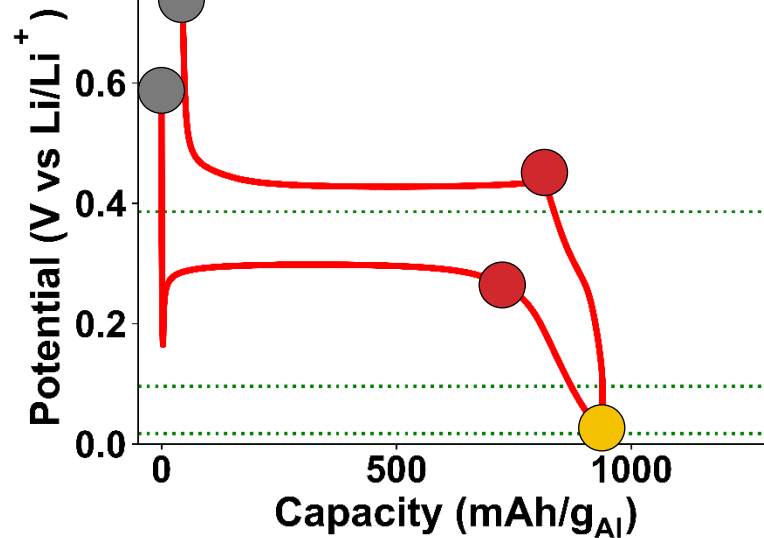
## Experimental Voltage Profile of Al-Li System

Theory

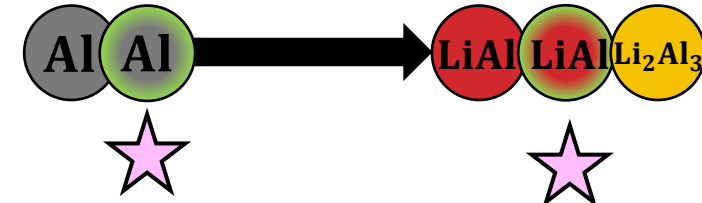
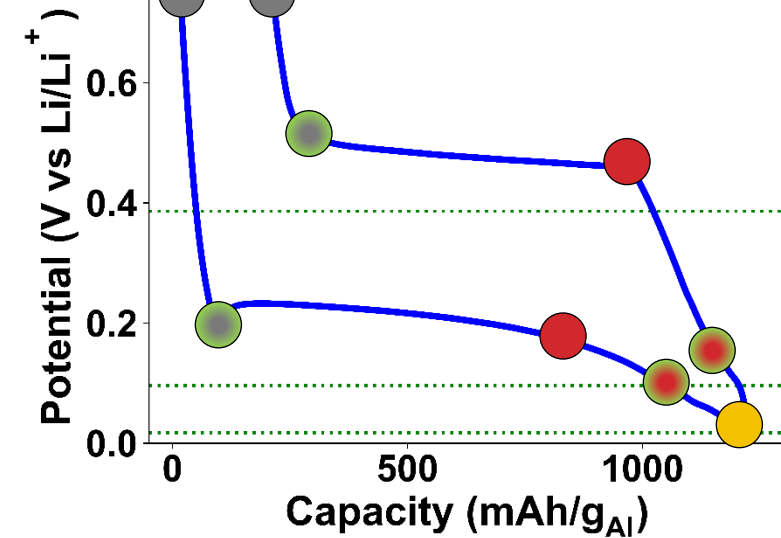
Theoretical Open  
Circuit Voltage  
(V vs Li)



Bulk Al



Nano Al



Nanostructure causes formation of metastable phase!

# (1) Phase Evolution

## Summary

### Structure

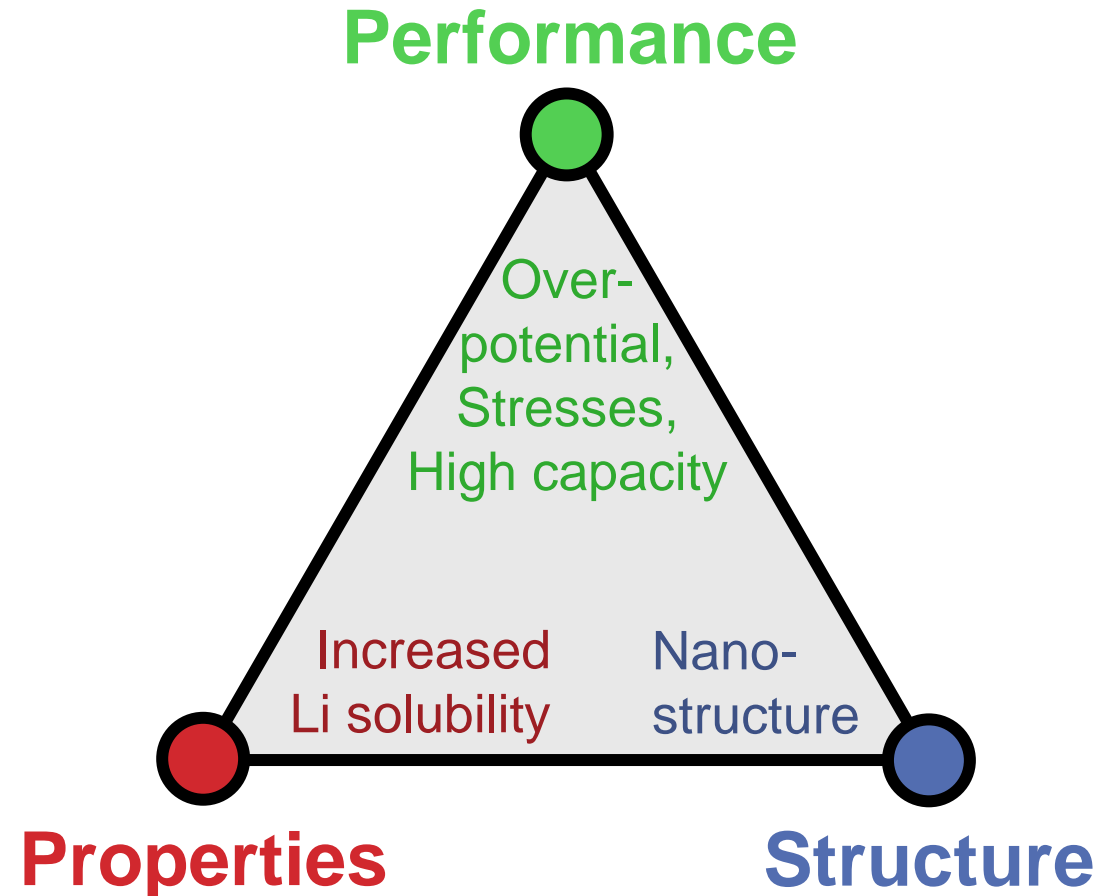
- Bulk vs. nanoporous anodes

### Properties

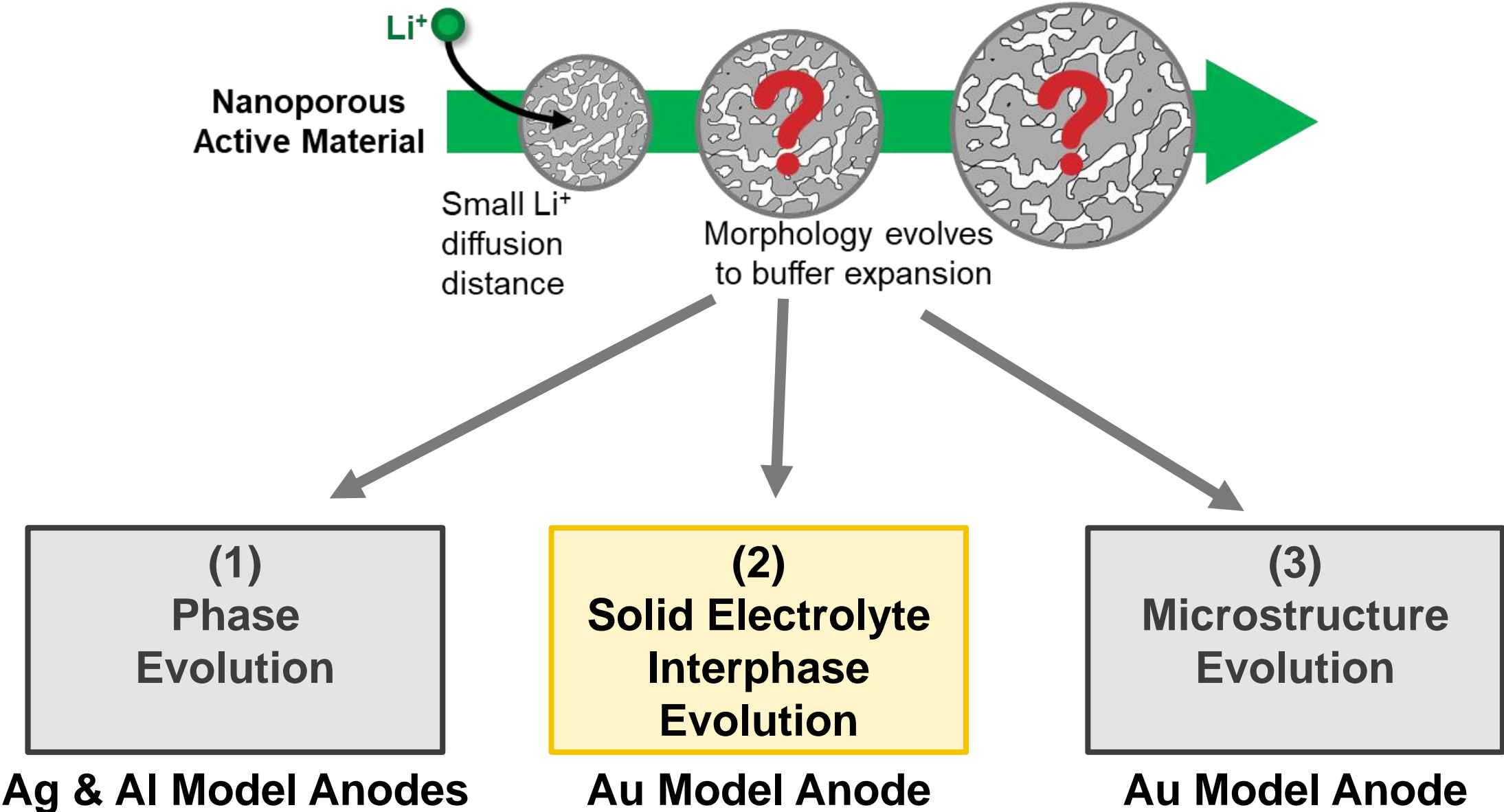
- Bulk anodes follow OCV simulation
- Nanoporous anodes have metastable phases

### Performance

- Nanoporous anodes have enhanced capacity
- Metastable phases cause overpotential, stresses

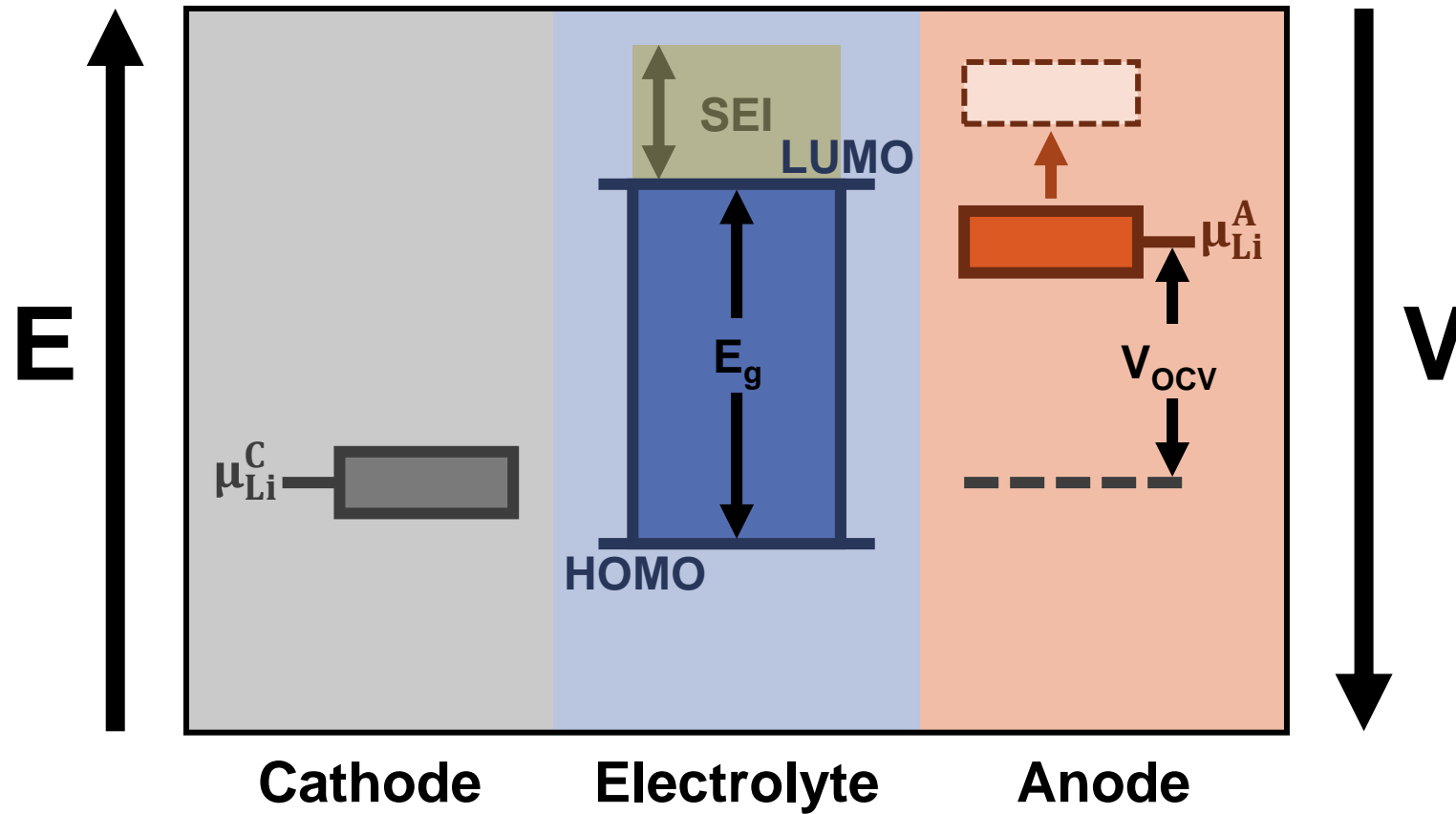


# Morphology Evolution During Reversible Li Storage



# (2) SEI Evolution

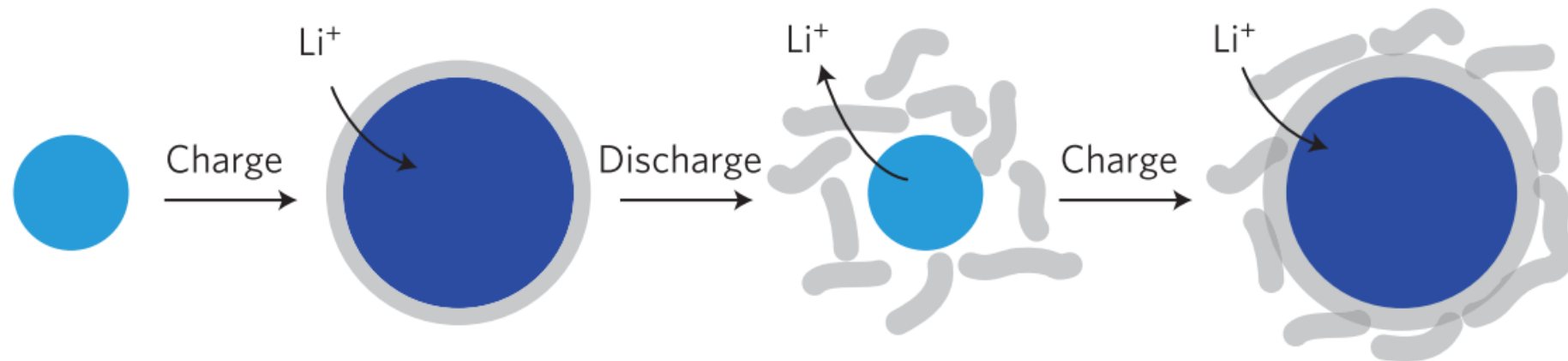
## Formation of solid electrolyte interphase (SEI)





## (2) SEI Evolution

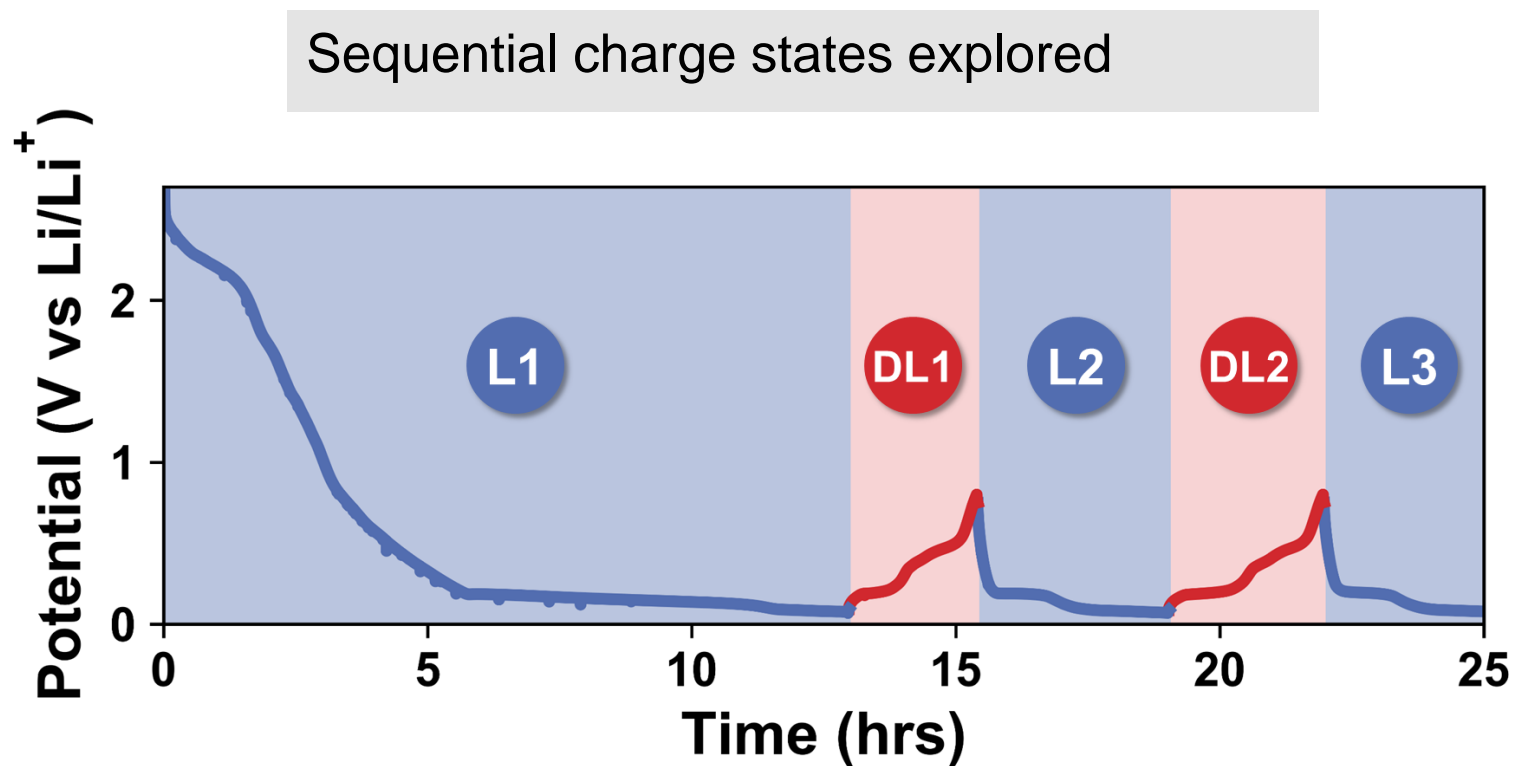
### Alloy anode SEI model for dense materials



**Key question:** Does this model apply to **nanoporous** anode materials?

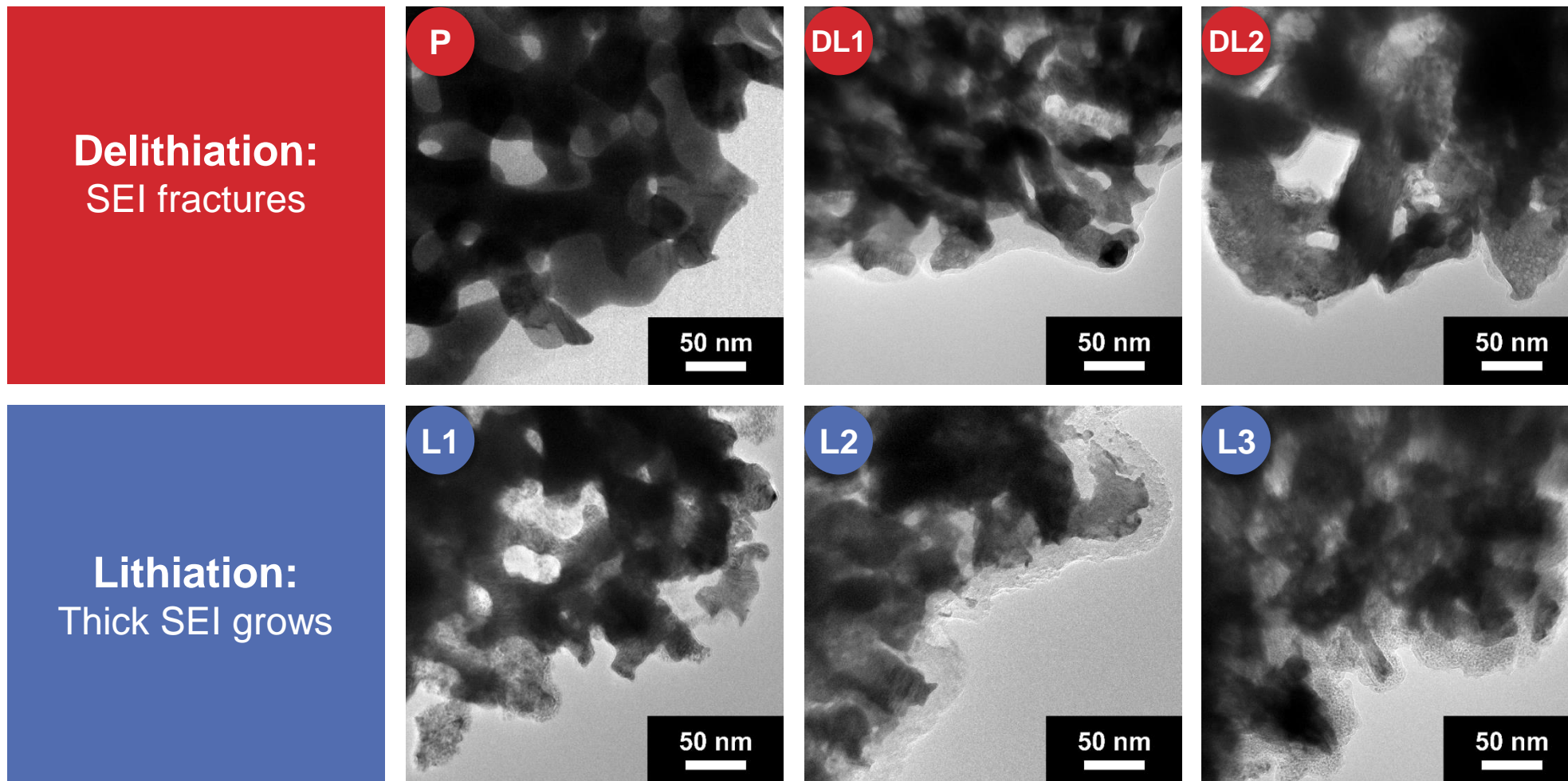
# (2) SEI Evolution

## Sequential Lithiation-Delithiation Cycling



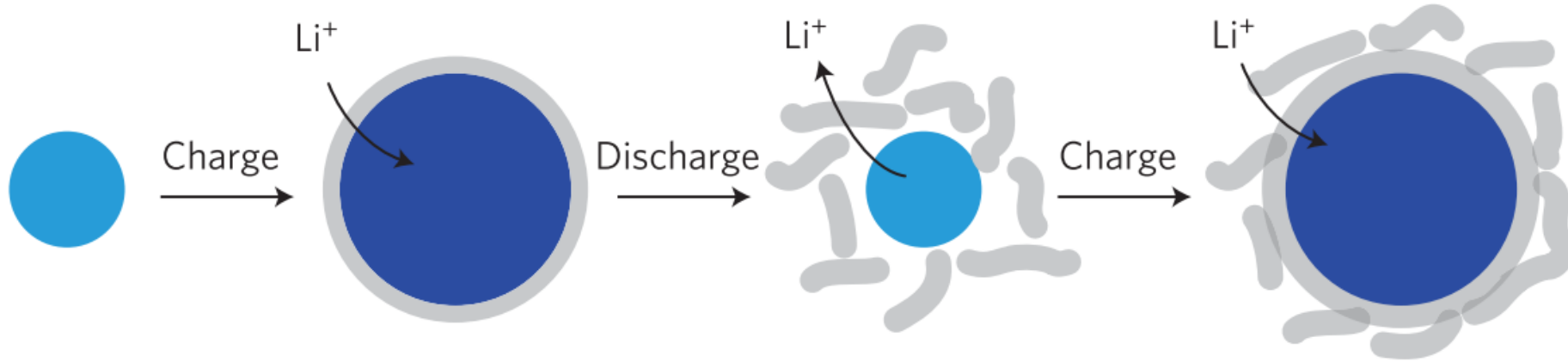
# (2) SEI Evolution

## TEM Characterization of SEI Evolution



# (2) SEI Evolution

## Conclusion

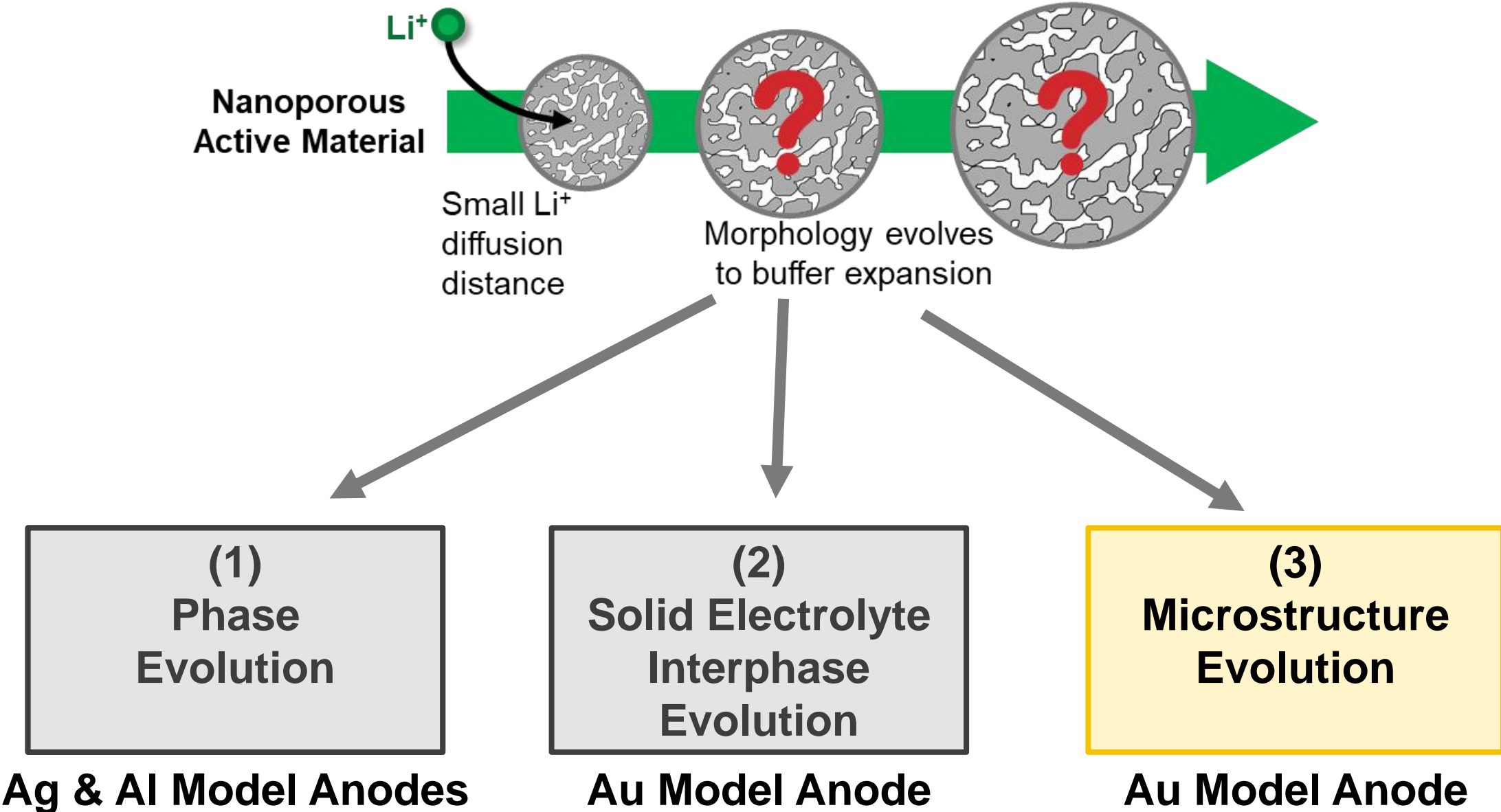


### **Model confirmed for nanoporous materials:**

- Thick SEI which grows in thickness with cycling
- SEI delamination during delithiation

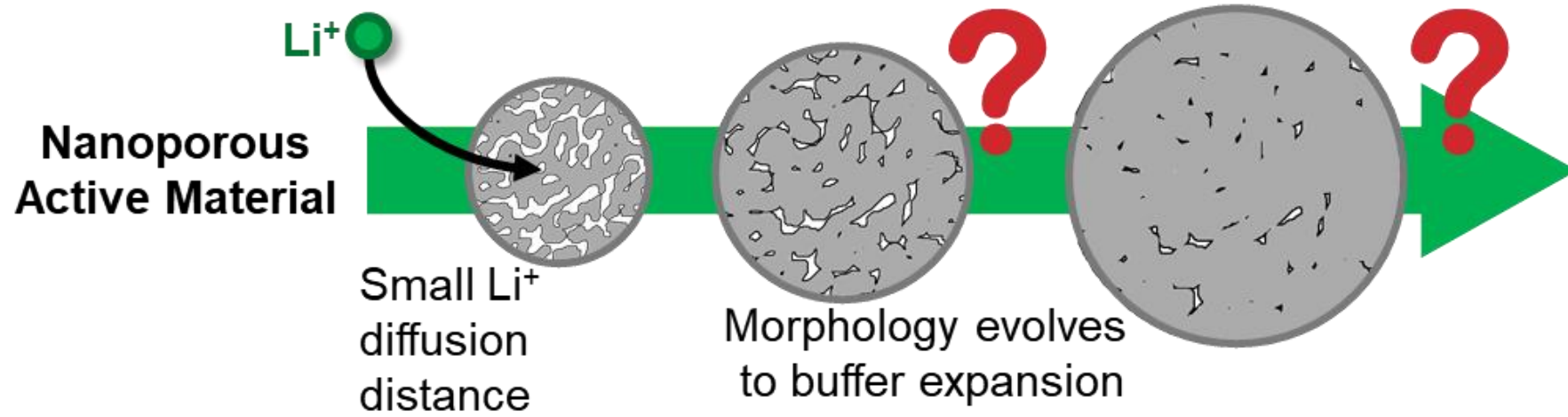
Dramatic SEI growth/fracture likely to impart stress on active material

# Morphology Evolution During Reversible Li Storage



# (3) Microstructure Evolution

## NP-Au as Model Anode for Investigation



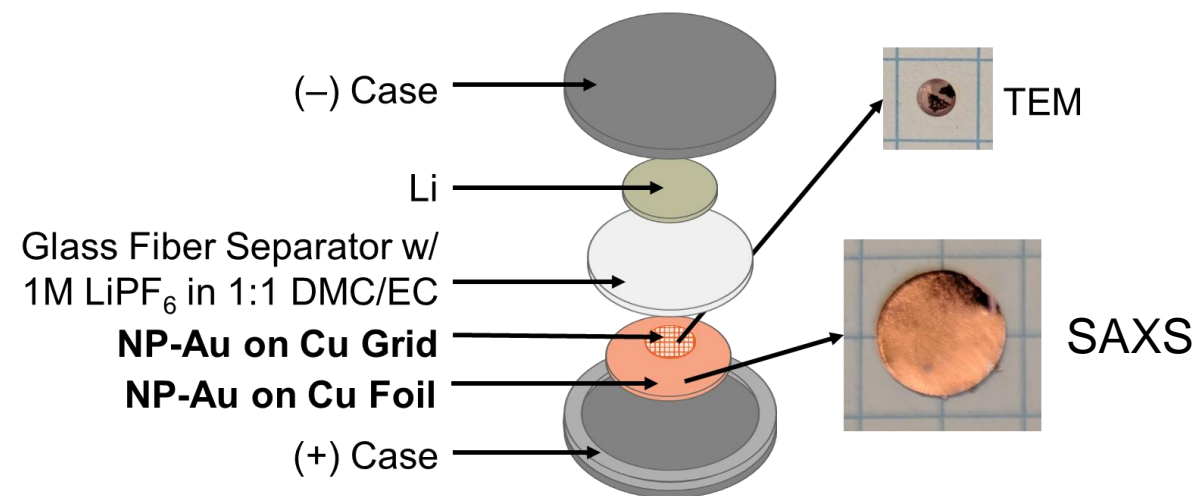
**NP-Au is used as a model anode for this investigation:**

- Model nanoporous metal
- Au has high contrast TEM images
- NP-Au has strong SAXS peak signature

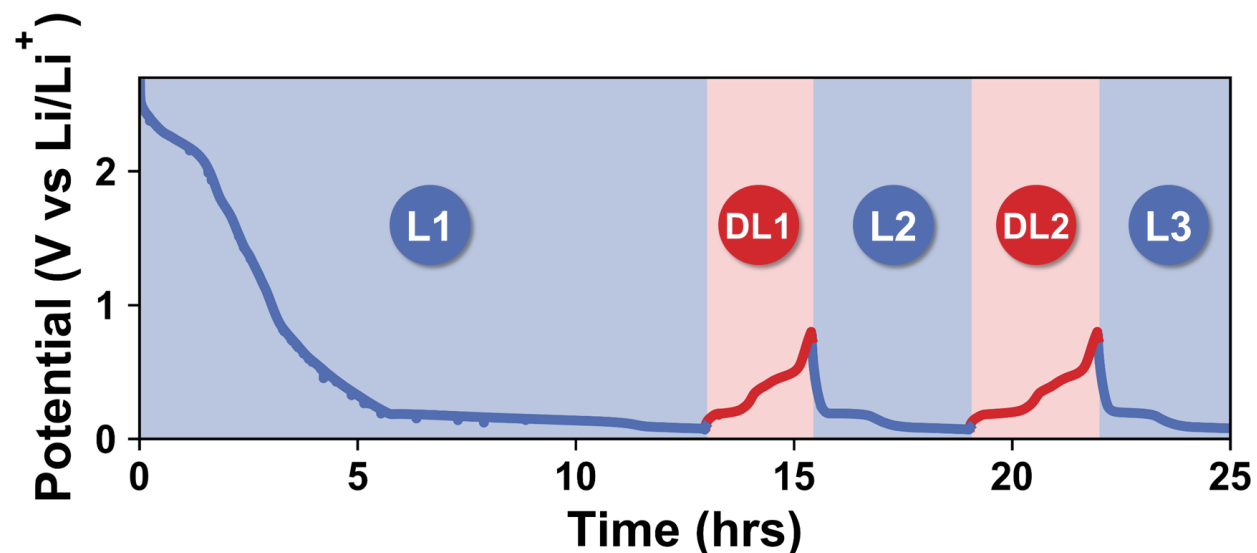
# (3) Microstructure Evolution

## Sequential Charge States Explored

Grid-in-a-coin cell configuration

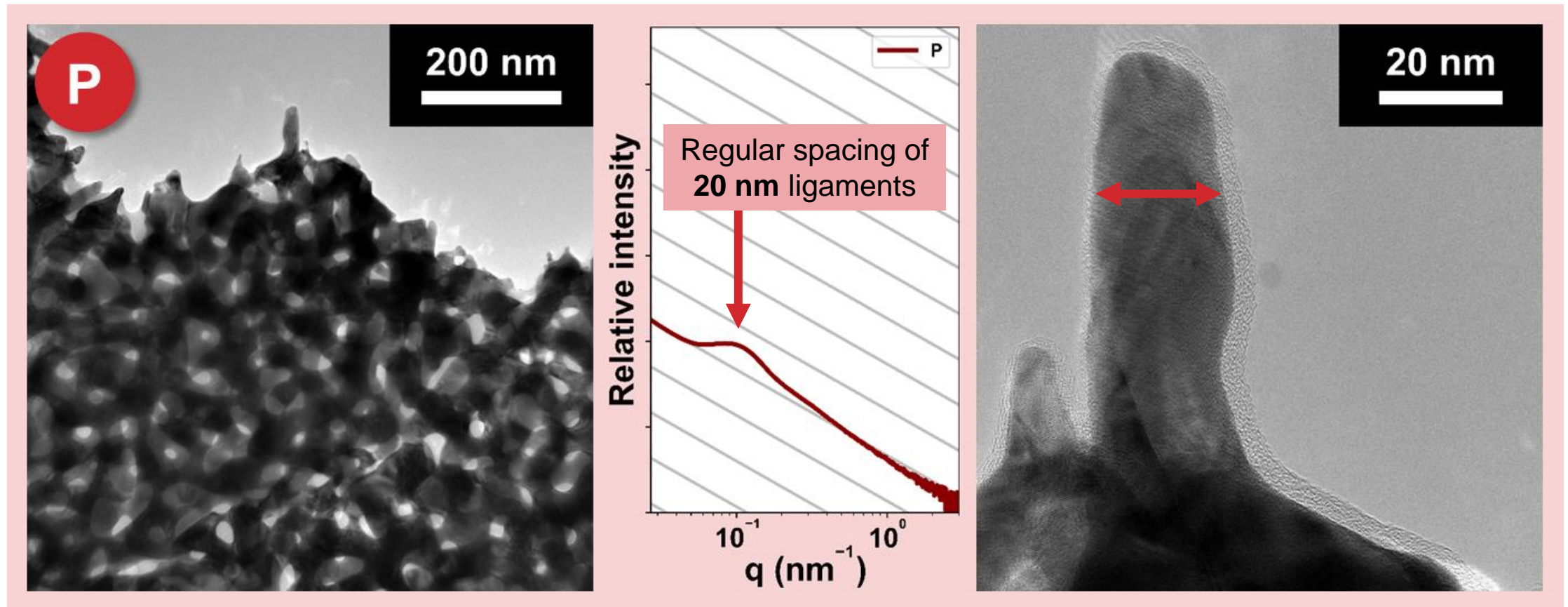


Sequential charge states explored



# (3) Microstructure Evolution

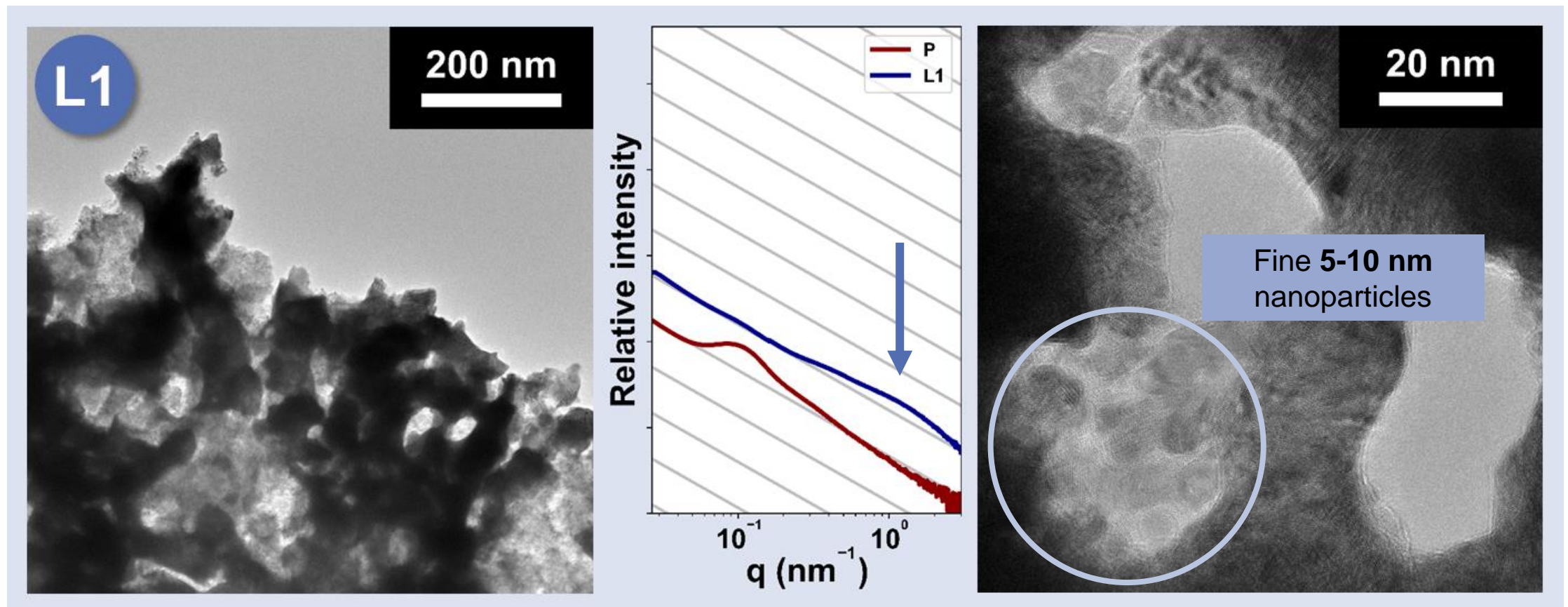
## Pristine (P)





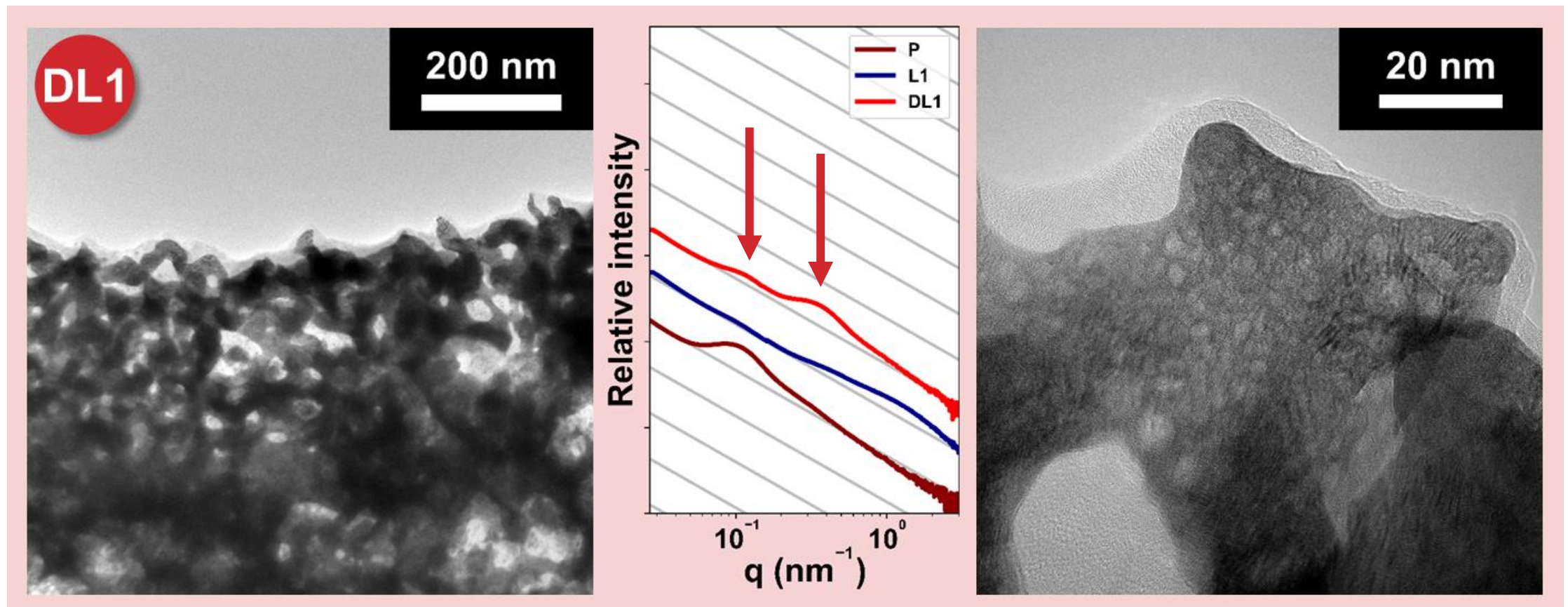
# (3) Microstructure Evolution

## First Lithiation (L1)



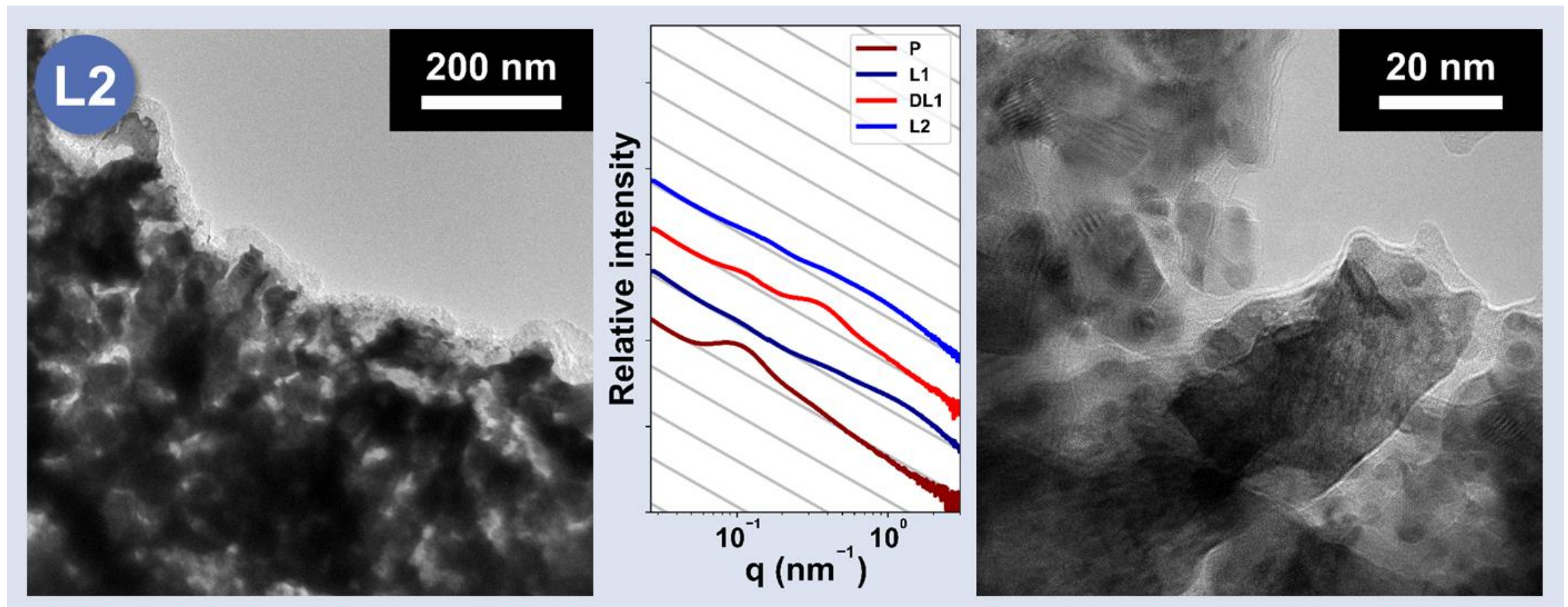
# (3) Microstructure Evolution

## First Delithiation (DL1)



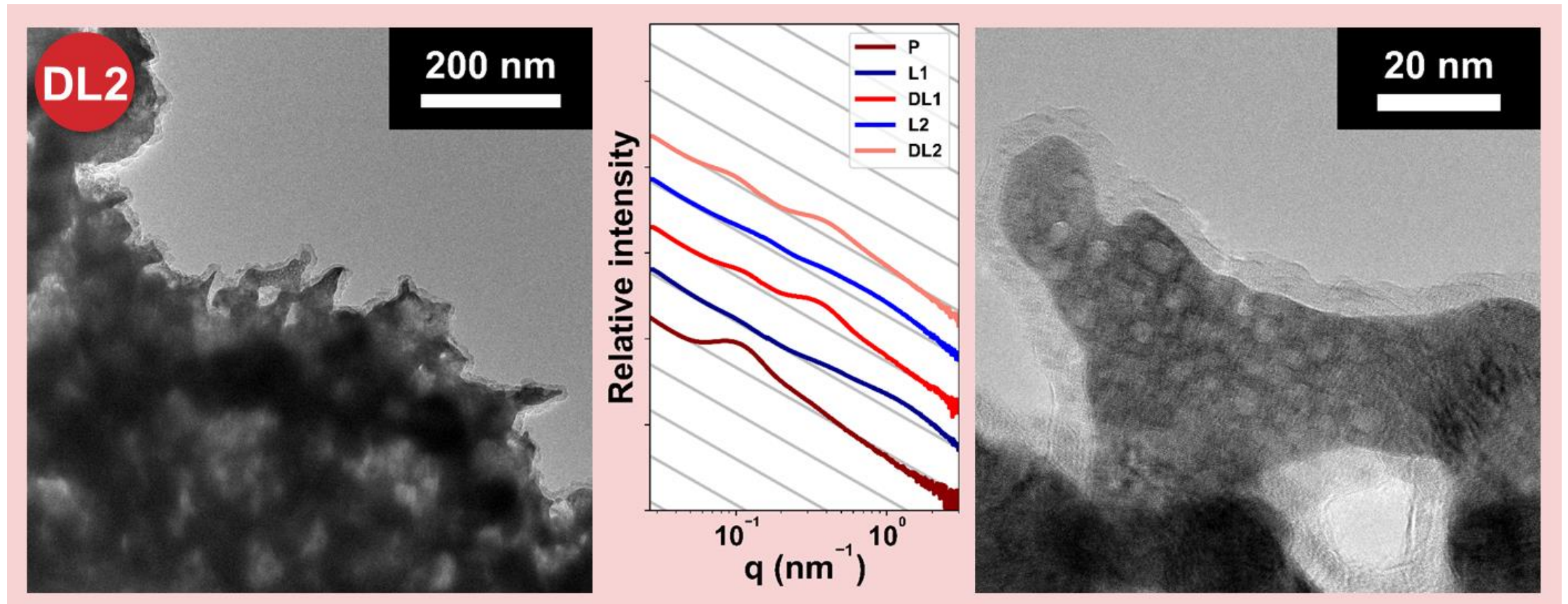
# (3) Microstructure Evolution

## Second Lithiation (L2)



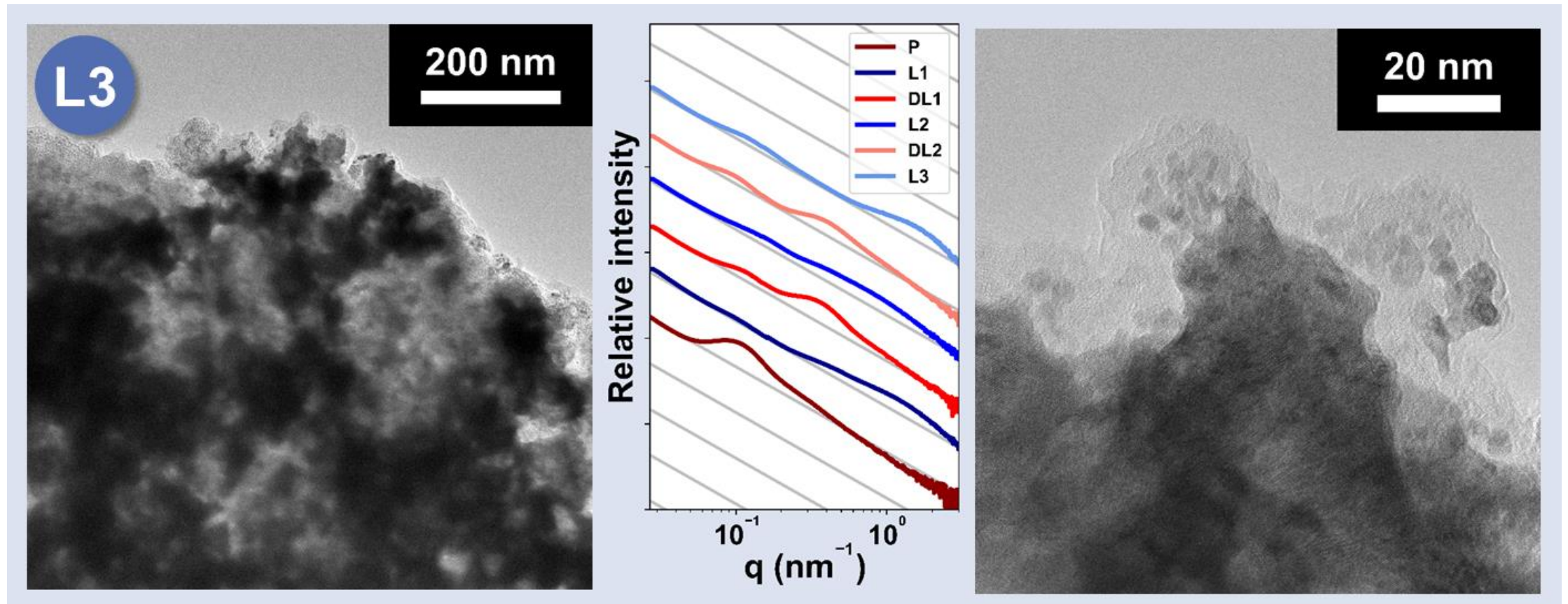
# (3) Microstructure Evolution

## Second Delithiation (DL2)



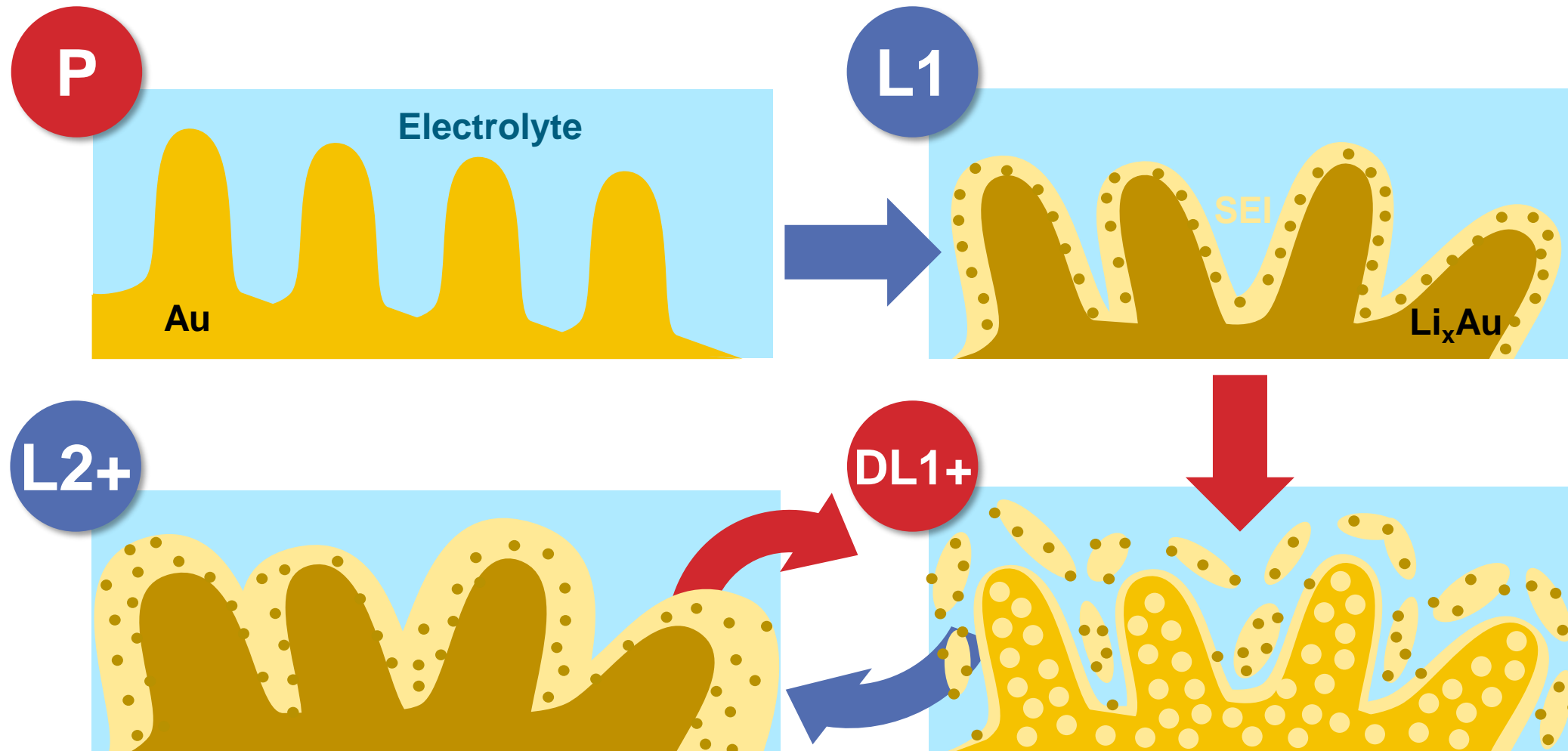
# (3) Microstructure Evolution

## Third Lithiation (L3)



# (3) Microstructure Evolution

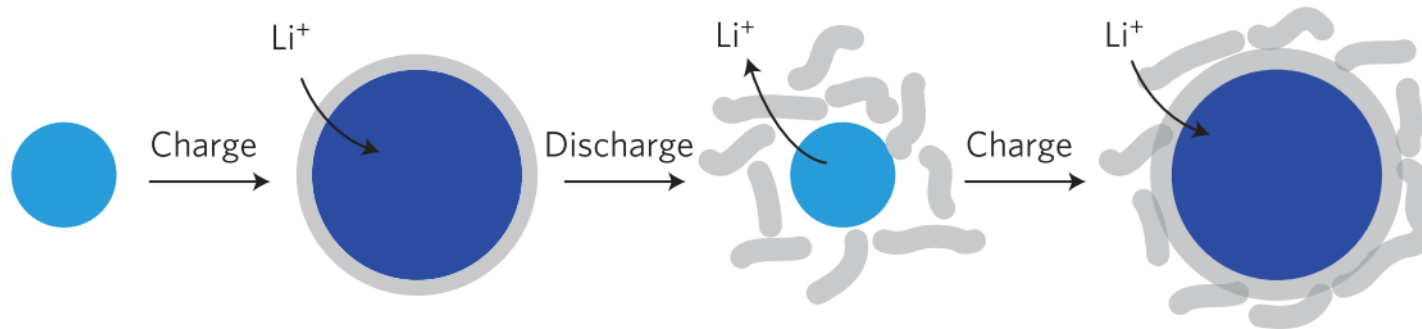
## Model: NP-Au Microstructural Evolution



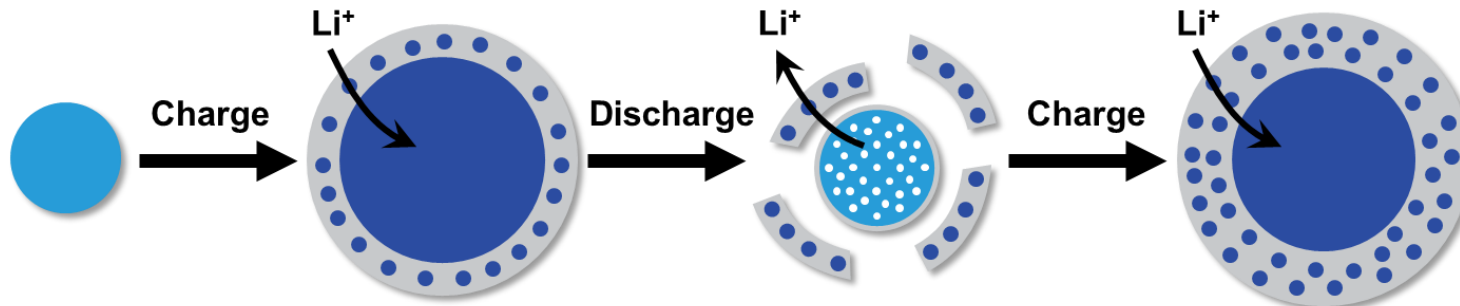
# (3) Microstructure Evolution

## Model: Updated Understanding of Alloy Anode Cycling

### Previous understanding



### New understanding from this work



# Conclusion

**What causes poor performance of nanoporous AI?**

**(1)  
Phase  
Evolution**

Stresses from metastable  
phases

**(2)  
Solid Electrolyte  
Interphase Evolution**

Stresses from dramatic SEI  
growth and removal  
processes

**(3)  
Microstructure  
Evolution**

Ligament degradation due to  
complex pulverization -  
dealloying mechanism



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**Thank you for your attention**